

pubs.acs.org/joc Article

# Catalytic *Syn*-Selective Nitroaldol Approach to Amphenicol Antibiotics: Evolution of a Unified Asymmetric Synthesis of (—)-Chloramphenicol, (—)-Azidamphenicol, (+)-Thiamphenicol, and (+)-Florfenicol

Yingqi Xia, Meifen Jiang, Minjie Liu, Yan Zhang, Hongmin Qu, Tong Xiong, Huashan Huang, Dang Cheng, and Fener Chen\*



Cite This: https://doi.org/10.1021/acs.joc.1c01124



**ACCESS** 

Metrics & More

Article Recommendations

Supporting Information

**ABSTRACT:** A unified strategy for an efficient and high diastereoand enantioselective synthesis of (—)-chloramphenicol, (—)-azidamphenicol, (+)-thiamphenicol, and (+)-florfenicol based on a key catalytic *syn*-selective Henry reaction is reported. The stereochemistry of the ligand-enabled copper(II)-catalyzed aryl aldehyde Henry reaction of nitroethanol was first explored to forge a challenging *syn*-2-amino-1,3-diol structure unit with vicinal stereocenters with excellent stereocontrol. Multistep continuous flow manipulations were carried out to achieve the efficient asymmetric synthesis of this family of amphenicol antibiotics.

$$R = NO_2, SO_2Me$$

$$R = NO_2, SO_2Me$$

$$HO$$

$$NO_2$$

$$R = NO_2, SO_2Me$$

$$R = NO_2, R' = OH, R' = R' = CI$$

$$(-)-\text{cohloramphenicol}, R = NO_2, R' = OH, R' = R' = CI$$

$$(-)-\text{azidamphenicol}, R = NO_2, R' = OH, R' = R' = CI$$

$$(+)-\text{thiamphenicol}, R = SO_2Me, R' = F, R' = R' = CI$$

### ■ INTRODUCTION

Amphenicols are a class of important synthetic antibiotics, which exhibit a broad spectrum of activity against both Gram-negative and Gram-positive microorganisms, such as *Streptococcus* spp., *Staphylococcus* spp., *Pasteurella* spp. etc. (-)-Chloramphenicol (1), (-)-azidamphenicol (2), (+)-thiamphenicol (3), and (+)-florfenicol (4) (Figure 1) are typical members of this family of antibiotics. They have been used effectively for the treatment of susceptible and serious bacterial infections in both human and livestock. Due to the lack of treatment options for multidrugresistant (MDR) bacterial infections with a global spread in the

Figure 1. Structure of amphenical antibiotics (1-4).

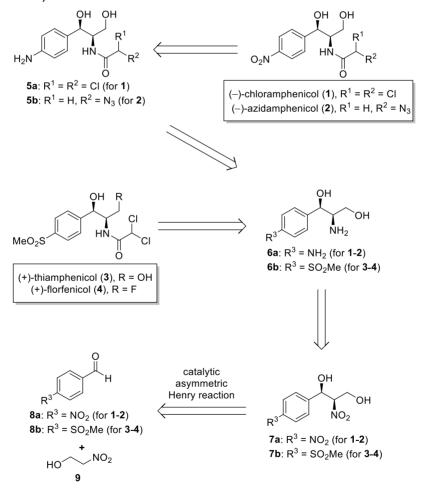
last few decades, which stand for a serious public health emergency, currently, this class of amphenical antibiotics has been reintroduced as antibacterial therapy for some infectious diseases, despite their undesirable side effects to patients. Once further evidence supports the broad clinical use of these older antibiotics for the bacterial infections caused by MDR microorganisms, the bulk supply of the active pharmaceutical ingredients for cost-effective and highly stereocontrolled manufacture of these amphenicol antibiotics would be an urgent concern. Structurally, these amphenical antibiotics (1-4) share a common chiral syn-2-amino-1,3-diol subunit, harboring two adjacent stereocenters, which presents a huge challenge to large-scale synthesis. Thus, this came as a driving force for the development of a variety of strategies for the construction of this highly functionalized syn-2-amino-1,3-diol moiety bearing (1R,2R)-vicinal stereocenters existing in this family of amphenicol antibiotics. To date, a number of asymmetric synthetic approaches toward the targeted amphenicol molecules have been reported.5 Typical methods include Corey's chiral diazaborolidine-mediated asymmetric aldol reaction, 5a Hajra's

Received: May 13, 2021



## Scheme 1. Previous Asymmetric Approaches Toward Amphenical Antibiotics (1-4)

Scheme 2. Retrosynthetic Analysis of 1-4



catalytic halohydroxylation, <sup>5b</sup> Wulff's catalytic asymmetric aziridination, <sup>5c</sup> Rao<sup>5d</sup> and Wu's<sup>5e</sup> catalytic Sharpless epoxidation, Dixon's silver-catalyzed isocyanoacetate aldol cyclization, <sup>5f</sup> Ratovelomanana-Vidal's Ruthenium, <sup>5g</sup> and Chen's <sup>5h</sup> enzymecatalyzed dynamic reductive kinetic resolution, Lin's enzymecatalyzed hydrocynation, <sup>5i</sup> etc (Scheme 1). However, synthetic

efficiency and stereoselectivities still need to be improved from the viewpoint of practical synthesis.

Our group has also developed several new synthetic strategies, including Sharpless<sup>6</sup> and vanadium-catalyzed<sup>7</sup> asymmetric epoxidation of allylic alcohol, Sharpless asymmetric dihydroxylation of acrylic esters,<sup>8</sup> ruthenium-catalyzed asymmetric transfer hydrogenation/dynamic kinetic resolution of *N*-Boc-

Table 1. Optimization of Reaction Conditions on the Copper-Catalyzed Asymmetric Henry Reaction of 8b with 9<sup>a</sup>

"Unless noted otherwise, reactions were performed with 8b (0.2 mmol), 9 (0.6 mmol),  $Cu(OAc)_2 \cdot H_2O$  (10 mol %), and L (10 mol %) in a solvent (0.8 mL) at 25 °C for 24 h. "Isolated yield." The *syn/anti* ratio and ee determined by chiral HPLC. "At 0 °C." At -5 °C. (4.0 equiv) in 0.3 mL THF. With  $Cu(OAc)_2 \cdot H_2O$  (5 mol %), L9 (5 mol %).

24

24

24

72

120

95

23

20

98

96

66:33

97:3

86:14

97:3

94:6

 $\alpha$ -amino- $\beta$ -ketoester, <sup>9</sup> thiourea-catalyzed asymmetric aldol reaction of aldehydes with isocyanatomalonate, <sup>10</sup> catalytic asymmetric Wulff-type aziridination of benzhydryl imine for

15

 $16^d$ 

 $17^e$ 

 $18^{d,f}$ 

 $19^{d,f,g}$ 

L9

L9

L9

L9

L9

**DMF** 

THF

THF

**THF** 

THF

the assembly of the *syn-*(1*R*,2*R*)-2-amino-1,3-diol framework, <sup>11</sup> and applied these techniques to the enantioselective synthesis of these amphenical antibiotics. However, a significant drawback

5

97

96

97

97

Cu(OAc)2 H2O (10 mol %)

ОН

OH

Table 2. Substrate Scope of Aryl Aldehydes

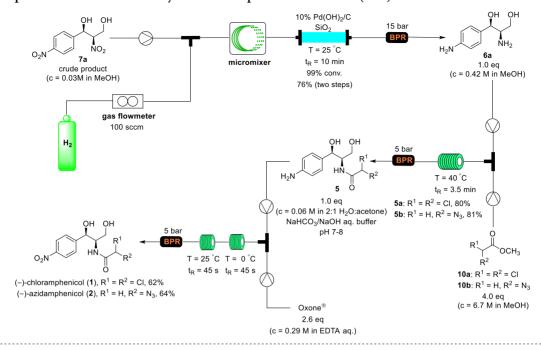
<sup>a</sup>Reaction conditions: 8 (1.0 mmol), 9 (4.0 mmol), Cu(OAc)<sub>2</sub>·H<sub>2</sub>O (10 mol %), and L9 (10 mol %) in THF (1.5 mL) at 0 °C for 72−120 h. The isolated yield was provided. The *syn/anti* ratio and the ee value were determined by chiral HPLC analysis. <sup>b</sup>Conversion yield determined by HPLC.

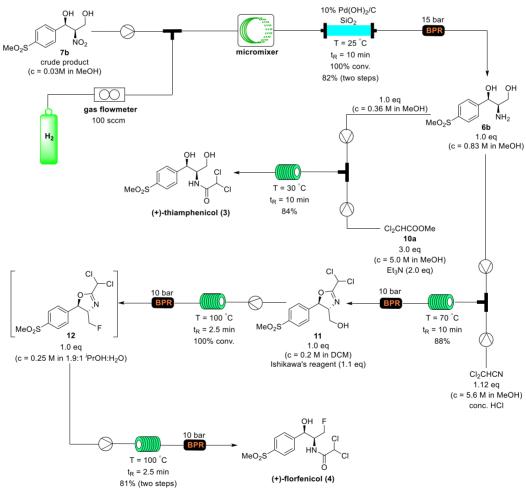
associated with these platforms is the synthetic difficulty in the stereochemical control of the *syn*-relationship between the *C*-1 and *C*-2 stereocenters during the construction of highly functionalized chiral scaffolds in their structures. To resolve this problem, it would be desirable to establish a practical and flexible strategy (i.e., short steps, facile reaction conditions, cheap commercially available starting materials, and excellent stereoselectivity) leading to the asymmetric synthesis of all of the amphenicol antibiotics and their analogues in this family bearing *syn*-2-amino-1,3-diol with vicinal stereocenters.

An attractive alternative, especially considering atom and step economy, <sup>12</sup> would be via Chen's ligand-controlled copper(II)-catalyzed diastereo- and enantioselective nitroaldol (Henry reaction) process that could open the opportunity for the construction of the *syn-2*-nitro-1,3-diol unit possessing vicinal

stereocenters. <sup>13</sup> Although Chen's method is limited to only aliphatic substrates, this elegant catalytic *syn*-selective enantioselective nitroaldol approach caught our attention. Had this been the case, a catalytic *syn*-selective enantioselective nitroaldol reaction <sup>14</sup> could be expected, allowing access to the *syn*-2-nitro-1,3-diol motif with two adjacent stereocenters required for the enantioselective synthesis of (—)-chloramphenicol (1), (—)-azidamphenicol (2), (+)-thiamphenicol (3), and (+)-florfenicol (4). Based on this idea, a retrosynthetic analysis of 1–4 is depicted in Scheme 2. We envisioned that *syn*-2-amino-1,3-diols 6a and 6b are the advanced intermediates for the enantioselective synthesis of amphenicol antibiotics 1–2 and 3–4, which can be accessed from the corresponding *syn*-2-nitro-1,3-diols 7a and 7b via catalytic hydrogenation under continuous flow conditions, respectively. To access compounds 7a and 7b, a

Scheme 3. Completion of Enantioselective Synthesis of Amphenical Antibiotics (1-4) in Continuous Flow





challenging *syn*-selective and enantioselective Henry reaction of 4-nitrobenzaldehyde (8a) or 4-methylsulfonyl benzaldehyde (8b) with 2-nitroethanol (9), three compounds being commercially available, would wait for us to achieve.

# ■ RESULTS AND DISCUSSION

Having designed the novel route to amphenical antibiotics (1-4), we first investigated the copper(II)-catalyzed asymmetric

Henry reaction of 4-methylsulfonyl benzaldehyde (8b) and 2nitroethanol (9), and the results are summarized in Table 1. The desired syn-adduct product 7b was obtained in 68% yield and with a moderate dr (syn/anti) of 73:27 and an excellent enantioselectivity of 97% ee when using 10 mol % of Cu(OAc)<sub>2</sub>. H<sub>2</sub>O, 10 mol % of noncyclic chiral amino alcohol ligand L1, commonly used in copper-catalyzed asymmetric Henry reactions, in THF at room temperature for 24 h (entry 1). Other ligands for the Cu(II) catalyst were then investigated, including the known  $^{13,15}$  1,2-nonsubstituted diphenyl  $\beta$ -amino alcohol ligands L2-L5 and our newly developed disubstituted-1,2-diphenyl amino alcohol ligands L6 and L7, tetrasubstituted-1,2-diphenyl amino alcohol ligand L8, and 1,2-di-biphenylsubstituted amino alcohol ligands L9-L11 (see the Supporting Information for details). The reaction proceeded smoothly with the use of pyrrolidine-derived ligand L2, giving a result similar to the case of using ligand L1 (76%, syn/anti = 75:25, 97% ee) (entry 2). The yield could be improved to 95% using piperidinederived ligand L3 in place of ligand L2, but a poor enantioselectivity was observed (entry 3). The ligand L4 bearing a morpholino group showed no reactivity under the reaction conditions (entry 4). The reaction with the ligand L5, derived from (R)-binaphthylazepine, was also less effective, providing syn-adduct 7b in only 36% yield with a poor enantioselectivity of 22% ee (entry 5). Subsequent investigations were focused on the pyrrolidine-derived chiral ligands L6-L11, which feature a 1,2di or tetrasubstituted diphenyl group and a 1,2-dibiphenyl group bearing different electronic and steric properties. It was found that all of the reactions using L6-L11 as the ligands produced syn-adduct 7b in good yields with good syn-selectivity and excellent enantioselectivity (entries 6-11). The reaction using L9 as the ligand showed the best result, leading to the desired product 7b in 81% yield with 80:20 dr (syn/anti) and 97% ee (entry 9). Further optimization of reaction conditions indicated that some reaction parameters, including solvent, temperature, the molar ratio of 8b to 9, concentration, and the catalyst loading amount, have an impact on this Henry reaction. The use of other solvents, such as MeCN, DCM, DCE, and DMF, could improve the yield (90-95%), but the poor diastereo- and enantioselectivities were observed (entries 12-15), and THF remained the best choice for this reaction. Notably, the diastereoselectivity was remarkably increased to 97:3 (syn/anti), albeit with 23% yield without affecting the enantioselectivity when the reaction proceeded at 0 °C (entry 16). Further decreasing the reaction temperature to -5 °C, slightly lower yield and diastereo- and enantioselectivities were observed (entry 17). Furthermore, increasing the loading of 9 to 4.0 equivalents relative to that of 8b and concentration from 0.25 M to 0.67 M under the identical reaction conditions (THF, 0 °C) led to a 98% yield of syn-adduct 7b with 97:3 dr (syn/anti) and 97% ee after prolonged reaction time (3 days, entry 18). Interestingly, when the catalyst loading was reduced to 5 mol %, the same result could be obtained (96% yield, 94:6 dr, 97% ee), albeit with an even longer reaction time (5 days, entry 19). With a systematic survey, the optimal reaction conditions were determined, as shown in entry 18 of Table 1.

With the optimized reaction conditions in hand, the substituted scope of the aryl aldehydes was then investigated, and the results are shown in Table 2. A variety of benzaldehydes (8a-8t) bearing monosubstituents at any position of the phenyl ring were initially surveyed. Both electron-withdrawing and donating groups were compatible with this transformation, and a diverse array of functional groups, such as halogen, methyl-

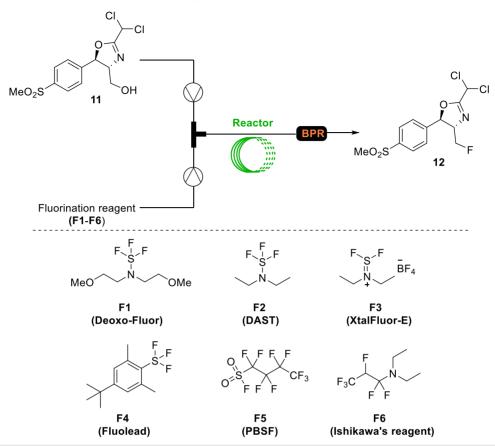
sulfonyl, nitro, trifluoromethyl, nitrile, ester, methoxy, and methylthio, were tolerated under the reaction conditions to give the desired products (7a-7t) in good to excellent yields (76-99%) with good to excellent diastereoselectivities (87:13 to 97:3 dr) and excellent enantioselectivities (95-99% ee). For benzaldehyde with disubstituents at any position of the phenyl ring, the reaction proceeded smoothly to generate 7u-7w in good to excellent yields (82-96%) with good to excellent diastereoselectivities (82:18 to 93:7 dr), and excellent enantioselectivities (95-99% ee). These results indicated that the electronic properties and steric hindrance exert little effect on this reaction. The trisubstituted benzaldehyde, 2,4,6trimethoxybenzaldehyde (8x), was also suitable for this reaction, delivering the desired syn-adduct product 7x in 98% yield with excellent diastereo- and enantioselectivity (99:1 dr, >99% ee). In addition to benzaldehydes, fused aryl (2-naphthyl) and fused heterocyclic aldehydes (8y, 8z, and 8aa) were also good substrates for this reaction providing the corresponding products (7y-7aa) with high efficiency (82-98% yield, 89:11 to 95:5 dr, 97-98% ee).

After the synthetic route to access the syn-2-nitro-1,3-diols has been established, we proceeded to develop multigram-scale continuous-flow processes for the enantioselective synthesis of amphenical antibiotics (1-4) (Scheme 3). 16 We began with syn-adduct 7a (95% ee) and syn-adduct 7b (97% ee) in our endeavor to develop the continuous-flow process. First, the continuous flow hydrogenation 17 of crude 7a was conducted in a fixed bed reactor (MF-200, Shenzhen E-Zheng Tech Co., Ltd.) containing 10% of Pd(OH)<sub>2</sub>/C dispersed in SiO<sub>2</sub> (5 mL internal volume) at 25 °C and 15 bar back-pressure, giving syn-2- amino-1,3-diol 6a in 76% yield in a residence time of 10 min. Remarkably, the catalytic system was found to be stable at room temperature for at least 10 days without reducing the catalytic activity. The syn-2-amino-1,3-diol 6a was combined with methyl dichloroacetate (10a) and methyl 2-azidoacetate (10b) pumped, respectively, into a 1.0 mL PTFE reactor coil (i.d. = 0.8 mm) at 40 °C with a residence time of 3.5 min to afford the corresponding 5a and 5b in 80% yield and 81% yield. The solution of 5a in H<sub>2</sub>O/acetone and aqueous NaHCO<sub>3</sub>/NaOH buffer was combined at a T-mixer with a stream of oxone in aqueous EDTA, and the oxidation reaction proceeded smoothly in a PTFE reactor coil with a residence time of 1.5 min to furnish (-)-chloramphenicol (1) in 62% yield after recrystallization from AcOEt/petroleum ether. It is noted that the flow protocol was optimized by operating the first half of the flow reactor at 0 °C, while the second half of the flow reactor at room temperature.

In addition to (-)-chloramphenicol (1), (-)-azidamphenicol (2) could also be efficiently prepared in 64% yield from **5b** under exactly the same reaction conditions using the same flow setup.

Next, we continued with continuous flow synthesis of (+)-thiamphenicol (3) and (+)-florfenicol (4). At the outset, catalytic hydrogenation of 7b delivered the desired product 6b in 82% yield under the same flow conditions. This flow reaction could be continuously operated for over 72 h without any clogging issues along with a throughput of 0.18 g·h<sup>-1</sup> (i.e., 4.3 g·day<sup>-1</sup>). The *syn-*2-amino-1,3-diol 6b could serve as a common intermediate for the asymmetric synthesis of target molecules (3 and 4) by subsequent transformations. A methanol solution of 6b, methyl dichloroacetate (10a), and Et<sub>3</sub>N was streamed through a PTFE reactor coil, and the acylation was found to proceed at 30 °C with 10 min residence time, affording (+)-thiamphenicol (3) in 84% yield.

Table 3. Optimization of the Fluorination Reaction Conditions in Flow



entry	conditions	$yield^b$ (%)
1	Deoxo-Fluor (2.0 equiv), DCM, 25 °C, 10 min	52
2	DAST (2.0 equiv), THF, 25 °C, 10 min	36
3	XtalFluor-E (2.0 equiv), DCM, 25 °C, 10 min	20
4	Fluolead (2.0 equiv), DCM, 25 °C, 10 min	24
5	PBSF (1.5 equiv), Et <sub>3</sub> N (3.0 equiv), Et <sub>3</sub> N·3HF (3.0 equiv), THF, 40 °C, 10 bar, 40 min	89
6	Ishikawa's reagent (1.1 equiv), DCM, 100 °C, 10 bar, 2.5 min	95
$a_{\text{Description}}$ 11 (10 mm) $a_{\text{Description}}$ 12 (10 mm) $a_{\text{Description}}$ 13 (10 mm) $a_{\text{Description}}$ 14 (10 mm)		

Reaction conditions: 11 (1.0 mmol) and solvent (5 mL) in flow. Isolated yield.

For the preparation of (+)-florfenicol (4), a solution of 6b in MeOH and a solution of dichloroacetonitrile in MeOH with a catalytic amount of conc. HCl were pumped separately into a reactor coil using two syringe pumps at 70 °C and 10 bar backpressure with a residence time of 10 min to afford the oxazolyl alcohol 11 in 88% yield. With access to the oxazolyl alcohol 11, the introduction of fluorine functionality to 11 in flow was at hand (Table 3). Initially, oxazolyl alcohol 11 was subjected to fluorination with Deoxo-Fluor (F1) into a reactor coil at 25 °C with 10 min residence time, which generated 52% yield of the desired oxazolyl fluoride 12, along with large amounts of unreacted 11 (entry 1). Further optimization studies were undertaken for this fluorination reaction. Other fluorination reagents (F2-F6) under reaction conditions illustrated in entries 2-6 of Table 3 were investigated. To our dismay, DAST (F2), XtalFluor-E (F3), and Fluolead (F4) only showed poor reactivity (entries 2-4). In contrast, PBSF (F5) performed better to provide the desired product 12 in 89% yield with a residence time of 40 min (entry 5). Further screening indicated that Ishikawa's reagent (F6) under flow conditions (100 °C, 10 bar back-pressure, and 2.5 min residence time) was the best fluorinating agent and cleanly produced the corresponding

product 12 in excellent yield (95%, entry 6). Finally, the crude product 12 was then dissolved in PrOH/H<sub>2</sub>O and subsequently streamed into another reactor coil at 100 °C and 10 bar backpressure. This flow protocol achieved 81% yield of (+)-florfenicol (4) in a residence time of 2.5 min.

#### CONCLUSIONS

In conclusion, we have successfully developed a unique and general strategy that allows the enantioselective synthesis of the amphenicol antibiotics, including (-)-chloramphenicol (1), (-)-azidamphenicol (2), and (+)-thiamphenicol (3) and its 3fluoro derivative, (+)-florfenicol (4), without the requirement for any protecting chemistry. The novel synthetic strategy was enabled by the key copper(II)-chiral biphenyl-substituted amino alcohol complex catalyzed Henry reaction, which allows efficient and high diastereo- and enantioselective construction of the common syn-2-amino-1,3-diol moiety. Our strategy can provide a platform for the straightforward and asymmetric synthesis of other pharmacologically important molecules and biologically active natural products containing the syn-2-amino-1,3-diol subunit with two vicinal stereocenters.

#### EXPERIMENTAL SECTION

General Information. All aldehydes (8a-8z and 8aa), 2nitroethanol (9), and solvents were commercially available. Amino alcohol ligands L1-L5 were prepared according to the literature procedures. 13,15 NMR data (1H, 13C, and 19F) were recorded on Bruker Avance III 600 and 400 spectrometers. Chemical shifts were referred to as TMS. HRMS were measured on a Bruker micro TOF Q III. Ultraviolet light (UV) detection was monitored at 254 nm. The procedure of continuous flow was monitored by thin-layer chromatography (TLC) and LC-MS (Agilent 6545 LC/Q-TOF, Agilent 1260 Infinity II, Eclipse Plus C18, RRHD 1.8  $\mu$ m, 2.1 × 50 mm<sup>2</sup>). Column chromatography was performed on silica gel (200-300 mesh), eluting with dichloromethane/methanol and petroleum ether/ethyl acetate. Melting points were measured on an SRS-optic melting point apparatus. In each case, syn/anti ratio and enantiomeric ratio were determined by HPLC analysis on a chiral column in comparison with authentic racemate, using a Daicel Chiralpak IC Column (250 × 4.6 mm<sup>2</sup>), Chiralpak AS-H Column (250  $\times$  4.6 mm<sup>2</sup>), or Chiralpak AD-H Column  $(250 \times 4.6 \text{ mm}^2)$ .

General Procedure for the Racemic Henry Reactions. To a mixture of aldehydes 8 (1 mmol, 1.0 equiv) and 2-nitroethanol 9 (3 mmol, 3.0 equiv) in THF (2 mL) was added TBAF (1 mL, 1.0 equiv, 1 M solution in THF) dropwise at room temperature and the mixture was stirred at room temperature for 36 h. Water (2 mL) was added, and the aqueous phase was extracted twice with ethyl acetate (2  $\times$  3 mL). The organic layer was combined and washed with brine (3 mL), then dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 100:2) to give the corresponding racemic products *rac-*7.

General Procedure for the Asymmetric Henry Reactions. A solution of  $\text{Cu}(\text{OAc})_2\cdot \text{H}_2\text{O}$  (0.1 mmol, 10 mol %) and L9 (0.1 mol, 10 mol %) in THF (1.5 mL) was stirred for 1 h at room temperature. Then, aldehyde 8 (1.0 mmol, 1.0 equiv) and 2-nitroethanol 9 (4.0 mmol, 4.0 equiv) were added into the reaction flask at 0 °C. The mixture was stirred for 72–120 h at 0 °C and then quenched with 1 M HCl (2 mL). The resulting mixture was extracted with ethyl acetate (3 × 10 mL). The organic layer was combined and washed with brine (5 mL). The organic phase was dried with Na<sub>2</sub>SO<sub>4</sub> and evaporated in a vacuum. The resulting residue was purified by silica gel chromatography (DCM/MeOH = 50:1) to give the corresponding *syn-2*-nitro-1,3-diols 7.

(1R,2R)-1-(4-(Methylsulfonyl)phenyl)-2-nitropropane-1,3-diol (7b). White solid, mp 136–138 °C, 60 h, 269.8 mg, 98% yield, syn/anti = 97:3; ¹H NMR (400 MHz, DMSO): δ 7.93 (d, J = 8.0 Hz, 2H), 7.71 (d, J = 8.4 Hz, 2H), 6.35 (d, J = 4.8 Hz, 1H), 5.28 (dd, J<sub>1</sub> = 6.4 Hz, J<sub>2</sub> = 4.4 Hz, 1H), 5.06 (dd, J<sub>1</sub> = 8.8 Hz, J<sub>2</sub> = 4.8 Hz, 1H), 4.80 (td, J<sub>1</sub> = 9.2 Hz, J<sub>2</sub> = 3.2 Hz, 1H), 3.84–3.77 (m, 1H), 3.25 (dt, J<sub>1</sub> = 12.4 Hz, J<sub>2</sub> = 4.0 Hz, 1H), 3.21 (s, 3H) ppm;  $^{13}$ C{ $^{1}$ H} NMR (100 MHz, DMSO): δ 146.5, 141.1, 128.4, 127.6, 95.4, 71.2, 60.5, 43.9 ppm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>6</sub>S + Na 298.0361, found 298.0362; [α]<sub>D</sub><sup>25</sup> = -32.6 (c = 0.29, EtOH). The enantiomeric excess of product 7b was determined by chiral HPLC: 97% ee (CHIRALPAK IC, hexane/i-PrOH = 65/35, flow rate: 0.7 mL/min, T = 30 °C, 210 nm), anti isomer: t<sub>major</sub> = 17.918 min, t<sub>minor</sub> = 20.068 min; syn isomer: t<sub>major</sub> = 30.064 min, t<sub>minor</sub> = 22.175 min.

(1R,2R)-1-(4-Fluorophenyl)-2-nitropropane-1,3-diol (7c). Colorless oil, 84 h, 96% yield, 206.6 mg, syn/anti = 94.6;  $^1H$  NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.46–7.38 (m, 2H), 7.15–7.05 (m, 2H), 5.02 (d, J = 7.2 Hz, 1H, anti), 5.01 (d, J = 9.2 Hz, 1H, syn), 4.79 (td,  $J_1 = 9.2$  Hz,  $J_2 = 3.2$  Hz, 1H), 4.18 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 9.2$  Hz, 1H, anti), 4.03 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 2.8$  Hz, 1H, anti), 3.82 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 9.2$  Hz, 1H, syn), 3.38 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 3.2$  Hz, 1H, syn) ppm;  $^{13}$ C{ $^1H$ } NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  162.8 (d, J = 244.4 Hz, syn), 162.7 (d, J = 243.8 Hz, anti), 135.9 (d, J = 3.2 Hz, anti), 135.7 (d, J = 3.1 Hz, syn), 128.6 (d, J = 8.2 Hz, syn), 128.3 (d, J = 243.8 Hz, anti), 115.1 (d, J = 21.8 Hz, syn), 114.8 (d, J = 21.8 Hz, anti), 95.1 (syn), 94.4 (anti), 71.6 (anti), 71.5 (syn), 60.4 (syn), 60.1 (anti) ppm. ESI HRMS: calcd. for C<sub>9</sub>H<sub>10</sub>FNO<sub>4</sub>+Na 238.0492, found 238.0497; [α]<sub>D</sub><sup>25</sup> = -30.9 (c = 0.61, EtOH). The enantiomeric excess of product 7c was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 95/5, flow

rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 22.550$  min,  $t_{\text{minor}} = 27.981 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 41.961 \text{ min}$ ,  $t_{\text{minor}} = 55.233 \text{ min}$ . (1R,2R)-1-(4-Chlorophenyl)-2-nitropropane-1,3-diol (7d). Pale yellow oil, 108 h, 93% yield, 215.4 mg, syn/anti = 94:6; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.40–7.35 (m, 4H), 5.02 (d, J = 9.2 Hz, 1H), 4.82-4.76 (m, 1H), 3.83 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 9.2$  Hz, 1H), 3.41 (dd,  $J_1$ = 12.4 Hz,  $I_2$  = 3.2 Hz, 1H) ppm;  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  139.6, 135.2, 129.7, 129.4, 96.1, 72.6, 61.5 ppm. ESI HRMS: calcd. for  $C_9H_{10}CINO_4+Na\ 254.0196$ , found 254.0192;  $[\alpha]_D^{25}=-21.9$  (c=1.76, EtOH). The enantiomeric excess of product 7d was determined by chiral HPLC: 98% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}}$  = 9.847 min,  $t_{\text{minor}} = 11.750 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 16.376 \text{ min}$ ,  $t_{\text{minor}} = 19.956 \text{ min}$ . (1R,2R)-1-(4-Bromophenyl)-2-nitropropane-1,3-diol (**7e**). Pale yellow oil, 108 h, 87% yield, 240.2 mg, syn/anti = 87:13; <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CD}_3\text{OD}): \delta 7.54 \text{ (d, } J = 8.4 \text{ Hz}, \text{2H, syn}), 7.50 \text{ (d, } J = 8.4 \text{ Hz}, \text{2Hz})$ 2H, anti), 7.33 (d, J = 8.4 Hz, 2H, syn), 7.30 (d, J = 8.4 Hz, 2H, anti),  $4.99 (d, I = 9.2 \text{ Hz}, 1\text{H}), 4.77 (td, I_1 = 9.6 \text{ Hz}, I_2 = 3.2 \text{ Hz}, 1\text{H}), 4.18 (dd, I_2 = 9.6 \text{ Hz}, I_3 = 9.6 \text{ Hz})$  $J_1 = 12.4 \text{ Hz}$ ,  $J_2 = 9.2 \text{ Hz}$ , 1H, anti), 3.99 (dd,  $J_1 = 12.0 \text{ Hz}$ ,  $J_2 = 2.8 \text{ Hz}$ , 1H, anti), 3.82 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 9.2 Hz, 1H, syn), 3.40 (dd,  $J_1$  = 12.0 Hz,  $J_2 = 3.2$  Hz, 1H, syn) ppm;  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  139.2 (anti), 138.9 (syn), 131.5 (syn), 131.2 (anti), 128.5 (syn), 128.2 (anti), 122.1 (syn), 121.8 (anti), 94.9 (syn), 94.2 (anti), 71.6 (anti), 71.5 (syn), 60.3 (syn), 59.9 (anti) ppm. ESI HRMS: calcd. for  $C_9H_{10}BrNO_4+Na$  297.9691, found 297.9695;  $[\alpha]_D^{25} = -15.3$  (c = 0.77, EtOH). The enantiomeric excess of product 7e was determined by chiral HPLC: 97% ee (CHIRALPAK IC, hexane/i-PrOH = 95/5, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 33.386$ min,  $t_{\text{minor}} = 41.268$  min; syn isomer:  $t_{\text{major}} = 64.533$  min,  $t_{\text{minor}} = 80.172$ 

(1R,2R)-2-Nitro-1-(4-(trifluoromethyl)phenyl)propane-1,3-diol (7f). Colorless oil, 84 h, 262.5 mg, 99% yield, syn/anti = 89:11; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.70 (d, J = 8.0 Hz, 2H, syn), 7.66 (d, J = 8.4 Hz, 2H, anti), 7.62 (d, J = 8.0 Hz, 2H, syn), 7.59 (d, J = 8.0 Hz, 2H, anti),  $5.13 (d, J = 9.2 Hz, 1H), 4.82 (td, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, 1H), 4.21 (dd, J_1 = 9.2 Hz, J_2 = 3.2 Hz, J_$  $J_1 = 12.4 \text{ Hz}, J_2 = 9.2 \text{ Hz}, 1\text{H}, anti), 4.00 (dd, J_1 = 12.4 \text{ Hz}, J_2 = 3.2 \text{ Hz},$ 1H, anti), 3.86 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 8.8$  Hz, 1H, syn), 3.45 (dd,  $J_1 =$ 12.4 Hz,  $J_2 = 3.2$  Hz, 1H, syn) ppm;  $^{13}C\{^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  142.9 (anti), 142.6 (syn), 128.8 (q, J = 32 Hz), 125.8 (syn), 125.4 (anti), 123.7 (q, J = 4 Hz, syn), 123.4 (q, J = 4 Hz, anti), 122.6 (q, J= 269 Hz), 93.2 (syn), 92.6 (anti), 70.0 (anti), 69.9 (syn), 58.7 (syn), 58.2 (anti) ppm. ESI HRMS: calcd. for  $C_{10}H_{10}F_3NO_4 + Na$  288.0460, found 288.0465;  $[\alpha]_D^{25} = -14.7$  (c = 2.32, EtOH). The enantiomeric excess of product 7f was determined by chiral HPLC: 97% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T= 30 °C, 220 nm), anti isomer:  $t_{\text{major}}$  = 6.902 min,  $t_{\text{minor}}$  = 7.896 min; syn isomer:  $t_{\text{major}} = 10.885 \text{ min}, t_{\text{minor}} = 12.577 \text{ min}.$ 

4-((1R,2R)-1,3-Dihydroxy-2-nitropropyl)benzonitrile (**7g**). White solid, mp 110–111 °C, 60 h, 217.8 mg, 98% yield, syn/anti = 92:8; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.77–7.74 (m, 2H), 7.63–7.61 (m, 2H), 5.14 (d, J = 8.8 Hz, 1H), 4.83–4.78 (m, 1H), 3.86 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 8.8$  Hz, 1H), 3.48 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 3.2$  Hz, 1H) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  143.8, 130.7, 126.1, 116.5, 110.5, 92.9, 69.8, 58.7 ppm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>4</sub> + Na 245.0538, found 245.0569; [α]<sub>D</sub><sup>25</sup> = -36.4 (c = 0.45, EtOH). The enantiomeric excess of product 7g was determined by chiral HPLC: 96% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 0.8 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{major} = 33.155$  min,  $t_{minor} = 35.837$  min; syn isomer:  $t_{major} = 59.506$  min,  $t_{minor} = 50.737$  min.

(1R,2R)-2-Nitro-1-phenylpropane-1,3-diol (7h). Pale yellow oil, 132 h, 167.6 mg, 85% yield, syn/anti = 94.6; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$ 7.41–7.36 (m, 5H), 5.46 (t, J = 4.4 Hz, 1H, anti), 5.24 (dd, J<sub>1</sub> = 8.8 Hz, J<sub>2</sub> = 3.2 Hz, 1H, syn), 4.81–4.76 (m, 1H, syn), 4.70–4.66 (m, 1H, anti), 4.25–4.07 (m, 2H, anti), 3.83–3.72 (m, 2H, syn), 3.21 (d, J = 4.8 Hz, 1H, anti), 2.85 (d, J = 4.0 Hz, 1H, syn), 2.61 (t, J = 7.2 Hz, 1H, anti), 2.28 (d, J = 6.0 Hz, 1H, syn) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  138.2 (anti), 138.0 (syn), 129.4 (syn), 129.2 (syn), 129.0 (anti), 128.9 (anti), 126.6 (syn), 125.9 (anti), 93.8 (syn), 91.9 (anti), 73.4 (anti), 72.6 (syn), 61.3 (syn), 60.0 (anti) ppm. ESI HRMS: calcd. for C<sub>9</sub>H<sub>11</sub>NO<sub>4</sub>+Na 220.0586, found 220.0582; [ $\alpha$ ]<sub>D</sub><sup>25</sup> = 2.0 (c = 0.79,

EtOH). The enantiomeric excess of product 7h was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 13.511$  min,  $t_{\text{minor}} = 15.645 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 20.455 \text{ min}$ ,  $t_{\text{minor}} = 27.789 \text{ min}$ . (1R,2R)-2-Nitro-1-(p-tolyl)propane-1,3-diol (7i). Pale yellow oil, 156 h, 173.2 mg, 82% yield, syn/anti = 88:12; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.29–7.16 (m, 4H), 4.99 (d, J = 7.2 Hz, 1H, anti), 4.94 (d, J= 9.6 Hz, 1H, syn), 4.80 (td,  $J_1$  = 9.6 Hz,  $J_2$  = 3.2 Hz, 1H), 4.20 (dd,  $J_1$  = 12.4 Hz,  $J_2 = 9.6$  Hz, 1H, anti), 4.05–4.00 (m, 1H, anti), 3.82 (dd,  $J_1 =$ 12.4 Hz,  $J_2 = 10.0$  Hz, 1H, syn), 3.34 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 2.8$  Hz, 1H, syn), 2.34 (s, 3H, syn), 2.32 (s, 3H, anti) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$ 138.5 (syn), 138.0 (anti), 136.8 (anti), 136.5 (syn), 129.1 (syn), 128.7 (anti), 126.5 (syn), 126.1 (anti), 95.3 (syn), 94.5 (anti), 72.2, 60.5 (syn), 60.1 (anti), 19.8 (syn), 19.8 (anti) ppm. ESI HRMS: calcd. for  $C_{10}H_{13}NO_4+Na$  234.0742, found 234.0738;  $[\alpha]_D^{25} =$ -11.4 (c = 0.56, EtOH). The enantiomeric excess of product 7i was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 14.779 \text{ min}$ ,  $t_{\text{minor}} = 18.756 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 18.756 \text{ min}$ 24.737 min,  $t_{\text{minor}} = 35.048$  min.

(1R,2R)-1-(4-Methoxyphenyl)-2-nitropropane-1,3-diol (7j). Colorless oil, 144 h, 184 mg, 81% yield, syn/anti = 93:7;  $^1$ H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.34–7.32 (m, 2H), 6.97–6.95 (m, 2H), 4.94 (d, J = 9.6 Hz, 1H), 4.84–4.78 (m, 1H), 3.83–3.79 (m, 4H), 3.36 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 3.2$  Hz, 1H) ppm;  $^{13}$ C{ $^1$ H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  161.3 (syn), 161.0 (anti), 132.9 (anti), 132.6 (syn), 129.0 (syn), 128.7 (anti), 115.0 (syn), 114.6 (anti), 96.6 (syn), 95.7 (anti), 73.2 (anti), 73.1 (syn), 61.7 (syn), 61.4 (anti), 56.0 (anti), 55.5 (syn) ppm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>5</sub>+Na 250.0691, found 250.0683; [ $\alpha$ ]<sub>D</sub><sup>25</sup> = -6.9 (c = 0.79, EtOH). The enantiomeric excess of product 7j was determined by chiral HPLC: 96% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{major} = 19.900$  min,  $t_{minor} = 25.022$  min; syn isomer:  $t_{major} = 36.504$  min,  $t_{minor} = 46.255$  min.

(1R,2R)-1-(4-(Methylthio)phenyl)-2-nitropropane-1,3-diol (7k). Pale yellow oil, 96 h, 228.7 mg, 94% yield,  $syn/anti = 89:11; {}^{1}H$  NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.35–7.23 (m, 4H), 5.01 (d, J = 7.2 Hz, 1H, anti), 4.96 (d, J = 9.6 Hz, 1H, syn), 4.84–4.78 (m, 1H), 4.20 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 9.2 Hz, 1H, anti), 4.03 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 3.2 Hz, 1H, anti), 3.83 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 9.2 Hz, 1H, syn), 3.39 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 3.2 Hz, 1H, syn), 2.48 (s, 3H, syn), 2.46 (s, 3H, syn) pm;  $^{13}C\{^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  139.7 (syn), 139.1 (syn), 136.4 (syn), 136.1 (syn), 127.1 (syn), 126.8 (syn), 126.1 (syn), 125.9 (syn), 94.4 (syn) pm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>4</sub>S+Na 266.0463, found 266.0465; [syn]) foliation (syn) pm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>4</sub>S+Na 266.0463, found 266.0465; [syn]) foliation (syn) remains equation (syn) and (syn) pm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>4</sub>S+Na 266.0463, found 266.0465; [syn]] for syn phase equation (syn) and (syn) pm. syn isomer: syn isomer:

(1R,2R)-1-([1,1'-Biphenyl]-4-yl)-2-nitropropane-1,3-diol (7l). White solid, mp 122–126 °C, 96 h, 259.6 mg, 95% yield, syn/anti = 93.7; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$ 7.58–7.49 (m, 4H), 7.38 (d, J = 8.0 Hz, 2H), 7.32 (t, J = 8.0 Hz, 2H), 7.26–7.21 (m, 1H), 4.95 (d, J = 9.2 Hz, 1H), 4.80–4.75 (m, 1H), 3.78 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 9.2$  Hz, 1H), 3.34 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 3.2$  Hz, 1H) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  141.5, 140.3, 138.6, 128.5, 127.2, 127.1, 127.0, 126.6, 95.2, 72.1, 60.5 ppm. ESI HRMS: calcd. for C<sub>15</sub>H<sub>15</sub>NO<sub>4</sub>+Na 296.0899, found 296.0894; [α]<sub>D</sub><sup>25</sup> = −24.3 (c = 0.68, EtOH). The enantiomeric excess of product 7l was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 85/15, flow rate: 0.8 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{major} = 12.060$  min,  $t_{minor} = 14.489$  min; syn isomer:  $t_{major} = 20.987$  min,  $t_{minor} = 23.151$  min.

4-((1R,2R)-1,3-Dihydroxy-2-nitropropyl)phenyl acetate (7m). Colorless oil, 132 h, 194.0 mg, 76% yield, syn/anti = 94:6; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.46–7.44 (m, 2H), 7.15–7.13 (m, 2H), 5.04 (d, J = 9.2 Hz, 1H), 4.84 (td, J<sub>1</sub> = 9.2 Hz, J<sub>2</sub> = 3.2 Hz, 1H), 3.85 (dd, J<sub>1</sub> = 12.4 Hz, J<sub>2</sub> = 9.6 Hz, 1H), 3.42 (dd, J<sub>1</sub> = 12.4 Hz, J<sub>2</sub> = 3.2 Hz, 1H), 2.27 (s, 3H) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  171.0, 152.1, 138.2, 128.8, 122.9, 96.2, 72.7, 61.5, 20.7 ppm. ESI HRMS: calcd. for

 $C_{11}H_{13}NO_6 + Na\ 278.0641$ , found 278.0640;  $[\alpha]_D^{25} = -17.6$  (c = 0.78, EtOH). The enantiomeric excess of product 7m was determined by chiral HPLC: 98% ee (CHIRALPAK IC, hexane/i-PrOH = 85/15, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}}$  = 15.340 min,  $t_{\text{minor}} = 17.904 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 24.200 \text{ min}$ ,  $t_{\text{minor}} = 28.347 \text{ min}$ . (1R,2R)-1-(2-Fluorophenyl)-2-nitropropane-1,3-diol (7n). Colorless oil, 72 h, 210.9 mg, 98% yield, syn/anti = 96:4; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.55 (td,  $J_1 = 7.6$  Hz,  $J_2 = 2.0$  Hz, 1H), 7.43–7.37 (m, 1H), 7.26 (t, J = 7.6 Hz, 1H), 7.17–7.13 (m, 1H), 5.36 (dd,  $J_1 = 9.2$  Hz,  $J_2 = 7.6$  Hz,  $J_3 = 7.6$  Hz,  $J_4 = 9.2$  Hz,  $J_5 = 9.2$  Hz,  $J_5$ 3.2 Hz, 1H), 4.96-4.93 (m, 1H), 3.96 (td,  $J_1 = 12$  Hz,  $J_2 = 2.8$  Hz, 1H), 3.96 (dt,  $J_1 = 12 \text{ Hz}$ ,  $J_2 = 3.6 \text{ Hz}$ , 1H) ppm;  $^{13}\text{C}\{^1\text{H}\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  160.9 (d, J = 244.0 Hz, syn), 160.9 (d, J = 244.1 Hz, anti), 131.3 (d, J = 8.5 Hz, syn), 131.0 (d, J = 7.7 Hz, anti), 129.3 (d, J = 3.9 Hz, syn), 129.2 (d, J = 3.8 Hz, anti), 127.6 (d, J = 13.2 Hz), 125.6 (d, J = 3.4 Hz, syn), 125.2 (d, J = 3.4 Hz, anti), 116.1 (d, J = 21.4 Hz, syn), 116.0 (d, *J* = 21.7 Hz, anti), 95.6 (d, *J* = 2.5 Hz, syn), 93.5 (anti), 67.6 (anti), 66.9 (syn), 61.2 (syn), 40.4 (anti) ppm. ESI HRMS: calcd. for  $C_9H_{10}FNO_4+Na$  238.0492, found 238.0487;  $[\alpha]_D^{25} = -12.0$  (c = 0.52, EtOH). The enantiomeric excess of product 7n was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 12.327$ min,  $t_{\text{minor}} = 15.067$  min; syn isomer:  $t_{\text{major}} = 16.038$  min,  $t_{\text{minor}} = 20.773$ 

(1R,2R)-1-(2-Chlorophenyl)-2-nitropropane-1,3-diol (70). Pale yellow oil, 108 h, 222.3 mg, 96% yield, syn/anti = 92:8; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.63 (dd,  $J_1 = 7.6$  Hz,  $J_2 = 2.0$  Hz, 1H), 7.47– 7.34 (m, 3H), 5.57 (d, J = 8.4 Hz, 1H), 4.99–4.94 (m, 1H), 4.14 (dd,  $J_1$ = 12.0 Hz,  $J_2$  = 9.6 Hz, 1H), 3.51 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 3.2 Hz, 1H) ppm;  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  138.2, 133.1, 130.6, 130.2, 129.2, 128.4, 96.0, 69.0, 61.1 ppm. ESI HRMS: calcd. for  $C_9H_{10}CINO_4+Na\ 254.0196$ , found 254.0193;  $[\alpha]_D^{25}=-11.4$  (c=1.15, EtOH). The enantiomeric excess of product 70 was determined by chiral HPLC: 97% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 15.711$  min,  $t_{\text{minor}} = 14.266 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 17.139 \text{ min}$ ,  $t_{\text{minor}} = 21.539 \text{ min}$ . (1R,2R)-1-(2-Bromophenyl)-2-nitropropane-1,3-diol (**7p**). White solid, mp 70–73 °C, 108 h, 270.6 mg, 98% yield, syn/anti = 92:8; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.61 (dd,  $J_1$  = 8.0 Hz,  $J_2$  = 1.2 Hz, 1H), 7.57 (dd,  $J_1 = 7.6$  Hz,  $J_2 = 2.0$  Hz, 1H), 7.43 (td,  $J_1 = 7.6$  Hz,  $J_2 = 1.2$  Hz, 1H), 7.25 (td,  $J_1 = 7.6$  Hz,  $J_2 = 2.0$  Hz, 1H), 5.49 (d, J = 8.4 Hz, 1H), 4.92-4.89 (m, 1H), 4.15-4.09 (m 1H), 3.43 (dd,  $J_1 = 12$  Hz,  $J_2 = 3.2$ Hz, 1H) ppm;  $^{13}$ C{ $^{1}$ H} NMR (100 MHz, CD $_{3}$ OD):  $\delta$  139.0, 132.6, 129.9, 128.5, 127.9, 122.1, 95.3, 70.4, 60.2 ppm. ESI HRMS: calcd. for  $C_9H_{10}BrNO_4+Na\ 297.9691$ , found 297.9689;  $[\alpha]_D^{25}=-3.3$  (c = 0.33, EtOH). The enantiomeric excess of product 7p was determined by chiral HPLC: 96% ee (CHIRALPAK IC, hexane/i-PrOH = 95/5, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 44.035$  min,  $t_{\text{minor}} = 37.405 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 48.952 \text{ min}$ ,  $t_{\text{minor}} = 65.461 \text{ min}$ . (1R,2R)-2-Nitro-1-(o-tolyl)propane-1,3-diol (**7q**). Colorless oil, 132 h, 171.1 mg, 81% yield, *syn/anti* = 91:9; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.48–7.45 (m, 1H, anti), 7.42–7.40 (m, 1H, syn), 7.27–7.14 (m, 3H), 5.34 (d, J = 6.8 Hz, 1H, anti), 5.29 (d, J = 9.6 Hz, 1H, syn), 4.93  $(td, J_1 = 9.6 \text{ Hz}, J_2 = 3.2 \text{ Hz}, 1\text{H}, syn), 4.28 (dd, J_1 = 12.8 \text{ Hz}, J_2 = 9.2 \text{ Hz},$ 1H, anti), 4.02 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 2.8 Hz, 1H, anti), 3.85 (dd,  $J_1$  = 12.0 Hz,  $J_2 = 9.6$  Hz, 1H, syn), 3.35 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 3.2$  Hz, 1H, syn), 2.42 (s, 3H, syn), 2.40 (s, 3H, anti) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD): δ 138.9 (anti), 138.6 (syn), 136.6 (syn), 136.3 (anti), 131.6 (syn), 131.4 (anti), 129.2 (syn), 129.0 (anti), 127.6 (syn), 127.4 (syn), 127.2 (anti), 127.0 (anti), 96.7 (syn), 93.9 (anti), 70.1 (anti), 69.6 (syn), 61.4 (syn), 60.8 (anti), 19.2 (syn), 18.8 (anti) ppm. ESI HRMS: calcd. for  $C_{10}H_{13}NO_4+Na\ 234.0742$ , found 234.0733;  $[\alpha]_D^{25} =$ -4.3 (c = 1.62, EtOH). The enantiomeric excess of product 7q was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 15.559 \text{ min}$ ,  $t_{\text{minor}} = 14.526 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 14.526 \text{ min}$ 

18.721 min,  $t_{\text{minor}} = 24.451$  min. (1R,2R)-1-(3-Chlorophenyl)-2-nitropropane-1,3-diol (7r). Colorless oil, 72 h, 224.7 mg, 97% yield, syn/anti = 95:5; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.48 (s, 1H), 7.40–7.34 (m, 3H), 5.10–5.08 (m, 1H, anti), 5.05 (dd,  $J_1 = 8.8$  Hz,  $J_2 = 2.4$  Hz, 1H, syn), 4.84–4.80 (m, 1H), 4.22 (td,

ı

 $J_1$  = 12.0 Hz,  $J_2$  = 2.4 Hz, 1H, anti), 4.07–4.01 (m, 1H, anti), 3.87 (td,  $J_1$  = 12.0 Hz,  $J_2$  = 2.4 Hz, 1H, syn), 3.47 (td,  $J_1$  = 12.0 Hz,  $J_2$  = 3.2 Hz, 1H, syn) ppm;  $^{13}$ C{ $^{1}$ H} NMR (100 MHz, CD<sub>3</sub>OD): δ 142.2 (anti), 142.0 (syn), 134.3 (syn), 134.1 (anti), 130.0 (syn), 129.7 (anti), 128.5 (syn), 128.1 (anti), 126.6 (syn), 126.3 (anti), 125.0 (syn), 124.7 (anti), 94.8 (syn), 94.1 (anti), 71.5 (anti), 71.5 (syn), 60.4 (syn), 59.9 (anti) ppm. ESI HRMS: calcd. for C<sub>9</sub>H<sub>10</sub>ClNO<sub>4</sub>+Na 254.0196, found 254.0195; [α]<sub>D</sub><sup>25</sup> = −12.1 (c = 1.86, EtOH). The enantiomeric excess of product 7r was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 95/5, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer: t<sub>major</sub> = 25.832 min, t<sub>minor</sub> = 30.432 min; syn isomer: t<sub>major</sub> = 46.861 min, t<sub>minor</sub> = 53.726 min.

(1R,2R)-1-(3-Bromophenyl)-2-nitropropane-1,3-diol (7s). Pale yellow oil, 84 h, 270.6 mg, 98% yield, syn/anti = 95:5; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.60 (t, J = 2.0 Hz, 1H), 7.52–7.49 (m, 1H), 7.38  $(dt, J_1 = 7.6 \text{ Hz}, J_2 = 1.6 \text{ Hz}, 1\text{H}), 7.30 (t, J = 8.0 \text{ Hz}, 1\text{H}), 5.03 (t, J = 7.2)$ Hz, 1H, anti), 5.00 (t, J = 9.2 Hz, 1H, syn), 4.77 (td,  $J_1 = 8.8$  Hz,  $J_2 = 3.2$ Hz, 1H), 4.18 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 9.2$  Hz, 1H, anti), 3.99 (dd,  $J_1 = 12.4$  Hz, 1H) 12.4 Hz,  $J_2$  = 2.8 Hz, 1H, anti), 3.83 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 9.2 Hz, 1H, syn), 3.43 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 3.2$  Hz, 1H, syn) ppm;  $^{13}C\{^1H\}$  NMR (100 MHz, CD<sub>3</sub>OD): δ 142.6 (anti), 142.3 (syn), 131.4 (syn), 131.0 (anti), 130.2 (syn), 129.9 (anti), 129.6 (syn), 129.3 (anti), 125.5 (syn), 125.1 (anti), 122.3 (syn), 122.0 (anti), 94.8 (syn), 94.1 (anti), 71.5 (anti), 71.4 (syn), 60.3 (syn), 59.8 (anti) ppm. ESI HRMS: calcd. for  $C_9H_{10}BrNO_4+Na\ 297.9691$ , found 297.9682;  $[\alpha]_D^{25} = -11.5$  (c = 1.81, EtOH). The enantiomeric excess of product 7s was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 12.247$  min,  $t_{\text{minor}} = 14.420 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 19.082 \text{ min}$ ,  $t_{\text{minor}} = 23.252 \text{ min}$ . (1R,2R)-2-Nitro-1-(m-tolyl)propane-1,3-diol (7t). Pale yellow oil, 132 h, 185.9 mg, 88% yield, syn/anti = 95:5; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.26 (t, J = 7.6 Hz, 1H), 7.22 (t, J = 2.0 Hz, 1H), 7.20–7.11 (m, 2H), 4.99 (d, J = 6.8 Hz, 1H, anti), 4.92 (d, J = 9.2 Hz, 1H, syn),4.82-4.76 (m, 1H), 4.19 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 9.2$  Hz, 1H, anti), 4.00 $(dd, J_1 = 12.4 \text{ Hz}, J_2 = 2.8 \text{ Hz}, 1\text{H}, anti), 3.81 (dd, J_1 = 12.4 \text{ Hz}, J_2 = 9.6)$ Hz, 1H, syn), 3.33 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 3.2 Hz, 1H, syn), 2.35 (s, 3H, syn), 2.33 (s, 3H, anti) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD): δ 140.8 (anti), 140.6 (syn), 139.5 (syn), 139.0 (anti), 130.2 (syn), 129.7 (anti), 129.4 (syn), 129.1 (anti), 128.2 (syn), 127.8 (anti), 124.7 (syn), 124.3 (anti), 96.4 (syn), 95.5 (anti), 73.5, 61.6 (syn), 61.0 (anti), 21.1 (anti), 21.1 (syn) ppm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>13</sub>NO<sub>4</sub>+Na 234.0742, found 234.0733;  $[\alpha]_D^{25} = -7.4$  (c = 0.75, EtOH). The enantiomeric excess of product 7t was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/ min, T = 30 °C, 220 nm), anti isomer:  $t_{\rm major}$  = 13.748 min,  $t_{\rm minor}$  = 17.288 min; syn isomer:  $t_{\text{major}} = 21.381$  min,  $t_{\text{minor}} = 26.903$  min.

(1R,2R)-1-(3,4-Dichlorophenyl)-2-nitropropane-1,3-diol (7u). Colorless oil, 96 h, 255.5 mg, 96% yield, *syn/anti* = 91:9; <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CD}_3\text{OD}): \delta 7.63 \text{ (d, } J = 2.0 \text{ Hz}, 1\text{H}, syn), 7.56 \text{ (d, } J = 8.4 \text{ Hz}, 1\text{Hz})$ 1H, syn), 7.52 (d, J = 8.4 Hz, 1H, anti), 7.37 (dd,  $J_1 = 8.4$  Hz,  $J_2 = 2.0$  Hz, 1H, syn), 7.32 (dd,  $J_1 = 8.4$  Hz,  $J_2 = 2.0$  Hz, 1H, anti), 5.07 (d, J = 8.8 Hz, 1H), 4.80 (td,  $J_1$  = 8.8 Hz,  $J_2$  = 3.2 Hz, 1H), 4.20 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 9.2 Hz, 1H, anti), 4.02 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 2.8 Hz, 1H, anti), 3.87 (dd,  $J_1 = 12.4 \text{ Hz}, J_{22} = 9.2 \text{ Hz}, 1\text{H}, syn), 3.51 (dd, J_1 = 12.0 \text{ Hz}, J_2 = 3.2 \text{ Hz}, 1\text{H}, syn) ppm. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD): <math>\delta$  140.8 (anti), 140.6 (syn), 132.3 (syn), 132.0 (syn), 131.8 (anti), 130.5 (syn), 130.2 (anti), 128.7 (syn), 128.4 (anti), 126.4 (syn), 126.1 (anti), 94.5 (syn), 94.0 (anti), 70.9 (anti), 70.8 (syn), 60.3 (syn), 59.8 (anti) ppm. ESI HRMS: calcd. for  $C_9H_9Cl_2NO_4+Na~287.9806$ , found 287.9808;  $[\alpha]_D^{25}$ = -22.9 (c = 1.75, EtOH). The enantiomeric excess of product  $7\mathbf{u}$  was determined by chiral HPLC: 97% ee (CHIRALPAK AS-H, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min,  $T = 30 \, ^{\circ}\text{C}$ , 220 nm), anti isomer:  $t_{\text{major}} = 30.273 \text{ min}, t_{\text{minor}} = 19.975 \text{ min}; syn \text{ isomer: } t_{\text{major}} =$ 27.313 min,  $t_{\text{minor}} = 41.672$  min.

(1*R*,2*R*)-1-(*Benzo*[*d*][1,3]*dioxol*-5-*yl*)-2-*nitropropane*-1,3-*diol* (**7v**). Yellow oil, 132 h, 197.8 mg, 82% yield, *syn/anti* = 82:18;  $^{1}$ H NMR (400 MHz, CD<sub>3</sub>OD): δ 6.91 (d, *J* = 1.6 Hz, 1H), 6.89–6.76 (m, 2H), 5.95 (s, 2H, *syn*), 5.94 (s, 2H, *anti*), 4.90 (d, *J* = 9.2 Hz, 1H), 4.79–4.73 (m, 1H), 4.17 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 9.2 Hz, 1H, *anti*), 4.04 (dd,  $J_1$  = 12.8 Hz,  $J_2$  = 3.2 Hz, 1H, *anti*), 3.81 (dd,  $J_1$  = 12.0 Hz,  $J_2$  = 9.6 Hz, 1H,

syn), 3.38 (dd,  $J_1 = 12.0 \text{ Hz}$ ,  $J_2 = 2.8 \text{ Hz}$ , 1H, syn) ppm;  ${}^{13}\text{C}\{{}^{1}\text{H}\}$  NMR (100 MHz, CD<sub>3</sub>OD): δ 148.2 (syn), 148.0 (syn), 147.8 (anti), 147.7 (anti), 133.7 (anti), 133.4 (syn), 120.3 (syn), 120.0 (anti), 107.8 (syn), 107.5 (anti), 106.4 (syn), 106.3 (anti), 101.2 (syn), 101.1 (anti), 95.3 (syn), 94.5 (anti), 72.1, 60.5 (syn), 60.2 (anti) ppm. ESI HRMS: calcd. for  $C_{10}H_{11}NO_6+Na\ 264.0484$ , found 264.0477;  $[\alpha]_D^{25}=-6.3$  (c=0.49, EtOH). The enantiomeric excess of product 7v was determined by chiral HPLC: 95% ee (CHIRALPAK IC, hexane/i-PrOH = 80/20, flow rate: 0.8 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 10.609$  min,  $t_{\text{minor}} = 14.754 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 17.166 \text{ min}$ ,  $t_{\text{minor}} = 18.520 \text{ min}$ . (1R,2R)-1-(2,3-Dimethoxyphenyl)-2-nitropropane-1,3-diol (7w). Colorless oil, 84 h, 239.2 mg, 93% yield, syn/anti = 93:7; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.12(t, J = 8.0 Hz, 1H), 7.05–6.99 (m, 2H), 5.34 (dd,  $J_1 = 9.2 \text{ Hz}$ ,  $J_2 = 1.6 \text{ Hz}$ , 1H), 4.93–4.88 (m, 1H), 3.96–3.92 (m, 1H), 3.88 (s, 3H), 3.87 (s, 3H), 3.36 (dt,  $J_1 = 12.0$  Hz,  $J_2 = 2.8$  Hz, 1H) ppm;  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  153.7 (syn), 153.6 (anti), 147.5 (syn), 147.2 (anti), 134.0 (anti), 133.9 (syn), 125.5 (syn), 124.9 (anti), 119.9 (syn), 119.8 (anti), 113.7 (syn), 113.6 (anti), 96.4 (syn), 93.8 (anti), 69.1 (anti), 67.8 (syn), 61.7 (syn), 61.1 (syn), 60.9 (anti), 60.2 (anti), 56.0 (syn) ppm. ESI HRMS: calcd. for  $C_{11}H_{15}NO_6+Na$  280.0797, found 280.0800;  $[\alpha]_D^{25} = 8.0$  (c = 1.13, EtOH). The enantiomeric excess of product 7w was determined by chiral HPLC: 99% ee (CHIRALPAK IC, hexane/i-PrOH = 80/20, flow rate: 0.8 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 15.954$  min,  $t_{\text{minor}} = 14.887 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 17.856 \text{ min}$ ,  $t_{\text{minor}} = 18.930 \text{ min}$ . (1R,2R)-2-Nitro-1-(2,4,6-trimet hoxyphenyl)propane-1,3-diol (7x). White solid, mp 160-163 °C, 84 h, 281.6 mg, 98% yield, syn/anti = 99:1; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  6.27 (s, 2H), 5.51 (d, J = 10.0Hz, 1H), 5.38 (td,  $J_1 = 10.4$  Hz,  $J_2 = 3.6$  Hz, 1H), 3.87 (s, 3H), 3.84 (s, 3H), 3.82-3.79 (m, 1H), 3.34-3.32 (m, 1H) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD): δ 162.2, 159.2, 106.1, 94.2, 90.6, 64.6, 60.7, 54.9, 54.5 ppm. ESI HRMS: calcd. for C<sub>12</sub>H<sub>17</sub>NO<sub>7</sub>+Na 310.0903, found 310.0910;  $[\alpha]_D^{25} = 15.4$  (c = 0.63, EtOH). The enantiomeric excess of product 7x was determined by chiral HPLC: >99% ee (CHIRALPAK AD-H, hexane/i-PrOH = 70/30, flow rate: 0.7 mL/min,  $T = 30 \,^{\circ}\text{C}$ , 220 nm), anti isomer:  $t_{\text{major}} = 10.902 \text{ min}, t_{\text{minor}} = 12.330 \text{ min}; syn \text{ isomer:}$ 

 $t_{\text{major}} = 15.695 \text{ min, } t_{\text{minor}} = 12.979 \text{ min.}$ (1R,2R)-1-(Naphthalen-2-yl)-2-nitropropane-1,3-diol (**7y**). Pale yellow oil, 132 h, 202.8 mg, 82% yield, syn/anti = 89:11; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.90–7.81 (m, 4H), 7.55–7.45 (m, 3H), 5.24  $(dd, J_1 = 7.2 \text{ Hz}, J_2 = 3.2 \text{ Hz}, 1\text{H}, anti), 5.18 (dd, J_1 = 9.6 \text{ Hz}, J_2 = 3.2 \text{ Hz},$ 1H, syn), 4.99-4.94 (m, 1H), 4.31-4.25 (m, 1H, anti), 4.10-4.05 (m, 1H, anti), 3.89 (td,  $J_1$  = 9.2 Hz,  $J_2$  = 2.4 Hz, 1H, syn), 3.38 (dt,  $J_1$  = 12.4 Hz,  $J_2$  = 2.8 Hz, 1H, syn) ppm; <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$ 137.3 (anti), 137.0 (syn), 133.6 (syn), 133.5 (anti), 133.4 (syn), 133.2 (anti), 128.5 (syn), 128.1 (anti), 127.8 (syn), 127.5 (syn), 127.4, 126.3 (syn), 126.2 (syn), 126.1 (anti), 126.1 (anti), 125.6 (anti), 123.8 (syn), 123.7 (anti), 95.3 (syn), 94.4 (anti), 72.7 (anti), 72.6 (syn), 60.7 (syn), 60.1 (anti) ppm. ESI HRMS: calcd. for C<sub>13</sub>H<sub>13</sub>NO<sub>4</sub>+Na 270.0742, found 270.0737;  $[\alpha]_D^{25} = -10.8$  (c = 0.68, EtOH). The enantiomeric excess of product 7y was determined by chiral HPLC: 98% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T= 30 °C, 220 nm), anti isomer:  $t_{\text{major}}$  = 16.384 min,  $t_{\text{minor}}$  = 21.644 min; *syn* isomer:  $t_{\text{major}} = 28.505 \text{ min}$ ,  $t_{\text{minor}} = 32.894 \text{ min}$ .

(1S,2R)-1-(Benzofuran-2-yl)-2-nitropropane-1,3-diol (7z). Yellow oil, 60 h, 232.5 mg, 98% yield, syn/anti = 90:10; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.58 (d, J = 7.6 Hz, 1H, syn), 7.54 (d, J = 7.6 Hz, 1H, anti), 7.48 (d, J = 8.4 Hz, 1H, syn), 7.45 (d, J = 8.0 Hz, 1H, anti), 7.29 (td,  $J_1 =$ 7.2 Hz,  $J_2 = 1.6$  Hz, 1H), 7.25–7.18 (m, 1H), 6.86 (s, 1H, syn), 6.79 (s, 1H, anti), 5.31 (d, J = 6.8 Hz, 1H, anti), 5.23 (d, J = 9.2 Hz, 1H, syn), 5.16-5.08 (m, 1H), 4.30 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 8.8$  Hz, 1H, anti), 4.11 $(dd, J_1 = 12.4 \text{ Hz}, J_2 = 1.6 \text{ Hz}, 1\text{H}, anti), 3.94 (dd, J_1 = 12.4 \text{ Hz}, J_2 = 9.2)$ Hz, 1H, syn), 3.60 (dd,  $J_1 = 12.4$  Hz,  $J_2 = 3.6$  Hz, 1H, syn) ppm;  ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CD<sub>3</sub>OD): δ 156.1 (anti), 156.0 (syn), 155.9 (anti), 155.2 (syn), 128.7 (anti), 128.6 (syn), 125.6 (syn), 125.4 (anti), 123.8 (syn), 123.7 (anti), 122.1 (syn), 121.9 (anti), 111.8 (syn), 111.7 (anti), 106.3 (syn), 105.3 (anti), 93.7 (syn), 92.4 (anti), 67.5 (anti), 66.8 (syn), 61.3 (syn), 60.8 (anti) ppm. ESI HRMS: calcd. for C<sub>11</sub>H<sub>11</sub>NO<sub>5</sub>+Na 260.0535, found 260.0532;  $[\alpha]_D^{25} = -39.9$  (c = 1.0, EtOH). The enantiomeric excess of product 7z was determined by chiral HPLC: 97% ee (CHIRALPAK IC, hexane/*i*-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), *anti* isomer:  $t_{\text{major}} = 11.790$  min,  $t_{\text{minor}} = 31.974$  min; *syn* isomer:  $t_{\text{major}} = 20.296$  min,  $t_{\text{minor}} = 24.369$  min.

(1R,2R)-1-(Benzo[b]thiophen-3-yl)-2-nitropropane-1,3-diol (7aa). Yellow oil, 96 h, 233.0 mg, 92% yield, syn/anti = 95:5; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  8.05 (d, J = 7.6 Hz, 1H, syn), 7.98 (d, J = 8.0 Hz, 1H, anti), 7.91-7.87 (m, 1H), 7.65 (s, 1H, syn), 7.59 (s, 1H, anti), 7.44-7.36 (m, 2H), 5.54 (d, J = 6.4 Hz, 1H, anti), 5.45 (d, J = 9.2 Hz, 1H, syn), 5.21–5.14 (m, 1H, syn), 5.12–5.07 (m, 1H, anti), 4.33 (dd, J<sub>1</sub> = 12.8 Hz,  $J_2$  = 9.6 Hz, 1H, anti), 4.07 (dd,  $J_1$  = 12.4 Hz,  $J_2$  = 2.8 Hz, 1H, anti), 3.92 (dd,  $J_1 = 12.4 \,\text{Hz}$ ,  $J_2 = 9.6 \,\text{Hz}$ , 1H, syn), 3.41 (dd,  $J_1 = 12.4 \,\text{Hz}$ ,  $J_2 = 3.6 \text{ Hz}, 1\text{H}, syn) \text{ ppm; } ^{13}\text{C}\{^1\text{H}\} \text{ NMR (100 MHz, CD}_3\text{OD): } \delta$ 141.9 (syn), 141.8 (anti), 137.9 (syn), 137.8 (anti), 135.8 (anti), 135.5 (syn), 126.0, 125.5 (syn), 125.4 (anti), 125.1 (syn), 125.0 (anti), 123.6 (syn), 123.6 (anti), 123.2 (syn), 122.7 (anti), 95.2 (syn), 93.7 (anti), 69.4 (anti), 68.6 (syn), 61.7 (syn), 60.8 (anti) ppm. ESI HRMS: calcd. for  $C_{11}H_{11}NO_4S+Na$  276.0306, found 276.0297;  $[\alpha]_D^{25} = -7.6$  (c =1.90, EtOH). The enantiomeric excess of product 7aa was determined by chiral HPLC: 98% ee (CHIRALPAK IC, hexane/i-PrOH = 90/10, flow rate: 1.0 mL/min, T = 30 °C, 220 nm), anti isomer:  $t_{\text{major}} = 15.083$ min,  $t_{\text{minor}} = 15.821$  min; syn isomer:  $t_{\text{major}} = 21.806$  min,  $t_{\text{minor}} = 26.908$ 

Continuous Flow Catalytic Hydrogenation Synthesis of 6a. (1R,2R)-2-Nitro-1-(4-nitrophenyl)propane-1,3-diol (7a). A solution of Cu(OAc)<sub>2</sub>·H<sub>2</sub>O (200 mg, 1.0 mmol, 10 mol %) and L9 (419.6 mg, 1.0 mmol, 10 mol %) in THF (15 mL) was stirred for 1 h at room temperature. Then, 4-nitrobenzaldehyde 8a (1.51 g, 10.0 mmol, 1 equiv) and 2-nitroethanol 9 (2.88 mL, 40.0 mmol, 4 equiv) were added into the reaction flask at 0  $^{\circ}$ C. The mixture was stirred for 72 h at 0  $^{\circ}$ C and then quenched with 1 M HCl (30 mL). The resulting mixture was extracted with ethyl acetate (3 × 60 mL). The organic layer was combined and washed with brine (3 × 50 mL). Then, the organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo to afford the crude syn-2-nitro-1,3-diol 7a, which was used in the next step without further purification. The crude product was analyzed by HPLC to determine the ratio of syn/anti and enantiomeric excess: syn/anti = 90:10, 95% ee (CHIRALPAK AD-H, hexane/i-PrOH = 80/20, flow rate: 0.8 mL/min, T = 30 °C, 210 nm), anti isomer:  $t_{\text{major}} = 16.378$  min,  $t_{\text{minor}} = 14.816 \text{ min}$ ; syn isomer:  $t_{\text{major}} = 17.846 \text{ min}$ ,  $t_{\text{minor}} = 28.347 \text{ min}$ .

(1R,2R)-2-Amino-1-(4-aminophenyl)propane-1,3-diol (6a). For the flow hydrogenation reaction, the flow system consists of a gas flow meter, a micropacked bed, and a 15 bar backpressure valve. The mixture containing the above crude 7a and methanol (333 mL) was delivered by a plunger pump (flow rate: 0.5 mL/min). H<sub>2</sub> gas from the high-pressure-resistant cylinder was fed via a gas flow meter (flow rate: 0.1 L/min). The liquid phase and gas phase were combined by a T-piece connector, entering a micromixer to control the reaction temperature at 25 °C. Then, the reaction stream was passed through a micropacked bed  $(2.0 \text{ g } 10\% \text{ Pd}(\text{OH})_2/\text{C}, 20.0 \text{ g})$ SiO<sub>2</sub>, and 5 mL internal volume) with 10.0 min residence time. The reaction stream was then collected in a 500 mL flat-bottomed conical flask and concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 3:1) to afford 6a (1.38 g, 76%, two steps) as a white solid. Mp 135-137 °C. <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  6.95 (d, J = 8.0 Hz, 2H), 6.51 (d, J = 8.0 Hz, 2H), 4.89 (s, 2H), 4.18 (d, J = 6.8 Hz, 1H), 3.39 (s, 4H), 3.22 (dd, J<sub>1</sub> = 10.4 Hz, J<sub>2</sub>)= 4.4 Hz, 1H), 3.07 (dd,  $J_1$  = 10.4 Hz,  $J_2$  = 6.4 Hz, 1H), 2.63 (td,  $J_1$  = 6.0 Hz,  $J_2 = 4.0$  Hz, 1H); <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO):  $\delta$  147.8, 131.5, 127.6, 113.8, 73.9, 63.4, 59.5 ppm. ESI HRMS: calcd. for  $C_9H_{14}N_2O_2$  + Na 205.0953, found 205.0956.

N-((1R,2R)-1-(4-Aminophenyl)-1,3-dihydroxypropan-2-yl)-2,2-di-chloroacetamide (5a). For the flow amidation reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume), a T-mixer, and a 5 bar back-pressure valve. Diamine 6a (1.82 g, 10.0 mmol, 1.0 equiv) was dissolved in degassed MeOH (24 mL) and placed in one syringe. The mixture containing methyl dichloroacetate 10a (5.72 g, 40.0 mmol, 4.0 equiv) and degassed MeOH (6 mL) was placed in another syringe. The diamine 6a syringe was set to  $205.2~\mu$ L/min, and the methyl dichloroacetate 10a syringe was set to  $80.8~\mu$ L/min. The flow was initiated, and the reaction mixture

was heated at 40 °C with a 3.5 min residence time. The reaction stream was collected in a 100 mL flat-bottomed conical flask. The reaction mixture was concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 100:5) to afford **5a** (2.35 g, 80%) as a yellow oil.  $^{1}$ H NMR (400 MHz, CD<sub>3</sub>OD): δ7.14 (d, J = 8.4 Hz, 2H), 6.71 (d, J = 8.4 Hz, 2H), 6.28 (s, 1H), 4.83 (s, 1H), 4.01 (q, J = 5.6 Hz, 1H), 3.69 (dd,  $J_1$  = 11.2 Hz,  $J_2$  = 6.0 Hz, 1H), 3.47 (dd,  $J_1$  = 11.2 Hz,  $J_2$  = 6.0 Hz, 1H);  $^{13}$ C{ $^{1}$ H} NMR (100 MHz, CD<sub>3</sub>OD): δ 166.3, 147.5, 132.8, 128.0, 116.3, 72.1, 67.4, 61.9, 58.9 ppm. ESI HRMS: calcd. for C<sub>11</sub>H<sub>14</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>3</sub> + Na 315.0279, found 315.0275, 317.0248, 319.2014.

Continuous Flow Synthesis of (-)-Chloramphenicol (1). For the flow oxidation reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume), a Tmixer, and a 5 bar back-pressure valve. 5a (2.0 g, 6.8 mmol, 1.0 equiv), NaHCO<sub>3</sub> (5.72 g, 68.0 mmol, 10.0 equiv), and NaOH (322 mg, 8.0 mmol, 1.18 equiv) were dissolved in degassed H<sub>2</sub>O/acetone (120 mL, v/v = 2:1) and placed in one syringe. The mixture containing oxone (10.86 g, 17.7 mmol, 2.6 equiv) and degassed  $4 \times 10^{-4}$  M EDTA aqueous (60 mL) was placed in another syringe. The 5a syringe was set to 429.1  $\mu$ L/min and the oxone syringe was set to 237.5  $\mu$ L/min. The flow was initiated with a 1.5 min residence time and cooling down the first half of the flow reactor to 0 °C, leaving the second half of the flow reactor length to warm up to room temperature. The reaction stream was collected in a 250 mL flat-bottomed conical flask and quenched with sodium bisulfite. The resulting mixture was extracted with ethyl acetate (3 × 100 mL). The organic layer was combined and washed with brine (100 mL). Then, the organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 10:1) to afford (-)-chloramphenicol (1) (1.58 g, 72%) as a white solid. The product was subsequently recrystallized from AcOEt/PE, affording (-)-chloramphenicol (1) (1.37 g, 62%). Mp = 150–151 °C;  $[\alpha]_D^{25}$  = -24.1 (c = 1.0, AcOEt); <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  8.33 (d, I =9.2 Hz, 1H), 8.17 (d, J = 8.8 Hz, 2H), 7.60 (d, J = 8.8 Hz, 2H), 6.48 (s, 1H), 6.05 (d, J = 4.4 Hz, 1H), 5.07 (dd,  $J_1 = 4.8$  Hz,  $J_2 = 2.4$  Hz, 1H), 5.00 (dd,  $J_1 = 6.0$  Hz,  $J_2 = 4.8$  Hz, 1H), 3.98–3.91 (m, 1H), 3.63–3.57 (m, 1H), 3.40–3.37 (m, 1H);  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, DMSO):  $\delta$ 163.9, 151.7, 146.9, 127.8, 123.3, 69.4, 66.9, 60.7, 57.3 ppm. ESI HRMS: calcd. for  $C_{11}H_{12}Cl_2N_2O_5$  + Na 345.0021, found 345.0010.

*N*-((1*R*,2*R*)-1-(4-Aminophenyl)-1,3-dihydroxypropan-2-yl)-2-azidoacetamide (**5b**). **5b** was prepared from **10b** in 81% yield by the same procedure of **5a**. Yellow oil. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.02 (d, *J* = 8.4 Hz, 2H), 6.60 (d, *J* = 8.4 Hz, 2H), 4.66 (d, *J* = 5.2 Hz, 1H), 3.96 (q, *J* = 6.0 Hz, 1H), 3.77 (s, 2H), 3.53 (dd, *J*<sub>1</sub> = 11.2 Hz, *J*<sub>2</sub> = 5.6 Hz, 1H), 3.34 (dd, *J*<sub>1</sub> = 11.2 Hz, *J*<sub>2</sub> = 6.0 Hz, 1H); <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  168.8, 146.8, 131.4, 126.9, 115.0, 71.5, 61.1, 57.2, 51.6 ppm. ESI HRMS: calcd. for C<sub>11</sub>H<sub>15</sub>N<sub>5</sub>O<sub>3</sub> + Na 288.1073, found 288.1069.

**Continuous Flow Synthesis of (–)-Azidamphenicol (2).** (–)-Azidamphenicol (2) was prepared from **5b** in 75% yield by the same procedure of (–)-chloramphenicol (1). The product was subsequently recrystallized from AcOEt/PE, affording (–)-azidamphenicol (2) in 64% yield. White solid. Mp = 108-109 °C;  $\left[\alpha\right]_{\rm D}^{25}$  = -9.8 (c = 1.0, EtOH);  $^{1}$ H NMR (400 MHz, DMSO):  $\delta$  8.17 (d, J = 8.8 Hz, 2H), 7.86 (d, J = 9.2 Hz, 1H), 7.60 (d, J = 8.8 Hz, 2H), 5.92 (d, J = 9.8 Hz, 1H), 5.04 (dd, 9.8 Hz, 1H), 4.03–3.97 (m, 1H), 3.71 (d, 9.8 Hz, 2H), 3.59–3.53 (m, 1H), 3.37–3.32 (m, 1H); 9.8 NMR (100 MHz, DMSO): 9.8 167.6, 152.1, 146.8, 127.7, 123.3, 69.7, 60.9, 56.7, 50.9 ppm. ESI HRMS: calcd. for 9.8 C<sub>11</sub>H<sub>13</sub>N<sub>3</sub>O<sub>5</sub> + Na 318.0814, found 318.0808.

Continuous Flow Catalytic Hydrogenation Synthesis of 6b. (1R,2R)-1-(4-(Methylsulfonyl)phenyl)-2-nitropropane-1,3-diol (7b). A solution of Cu(OAc) $_2$ ·H $_2$ O (200 mg, 1.0 mmol, 10 mol %) and L9 (419.6 mg, 1.0 mmol, 10 mol %) in THF (15 mL) was stirred for 1 h at room temperature. Then, 4-methylsulfonyl benzaldehyde 8b (1.84 g, 10.0 mmol, 1 equiv) and 2-nitroethanol 9 (2.88 mL, 40.0 mmol, 4 equiv) were added into the reaction flask at 0 °C. The mixture was stirred for 72 h at 0 °C, then quenched with 1 M HCl (30 mL). The resulting mixture was extracted with ethyl acetate (3 × 60 mL). The

organic layer was combined and washed with brine ( $3 \times 50 \text{ mL}$ ). Then, the organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo to afford the crude *syn*-2-nitro-1,3-diol 7b, which was used in the next step without further purification. The crude product was analyzed by HPLC to determine the ratio of *syn/anti* and enantiomeric excess: *syn/anti* = 95:5, 97% ee (CHIRALPAK IC, hexane/*i*-PrOH = 65/35, flow rate: 0.7 mL/min, T = 30 °C, 210 nm).

1R,2R)-2-Amino-1-(4-(methylsulfonyl)phenyl)propane-1,3-diol (6b). For the flow hydrogenation reaction, the flow system consists of a gas flow meter, a micromixer, a micropacked bed, and a 15 bar backpressure valve. The mixture containing the above crude 7b and methanol (333 mL) was delivered by a plunger pump (flow rate: 0.5 mL/min). H<sub>2</sub> gas from the high-pressure-resistant cylinder was fed via a gas flow meter (flow rate: 0.1 L/min). The liquid phase and gas phase were combined by a T-piece connector, entering a micromixer to control the reaction temperature at 25 °C. Then, the reaction stream was passed through a micropacked bed (2.0 g 10% Pd(OH)<sub>2</sub>/C, 20.0 g SiO<sub>2</sub>, and 5 mL internal volume) with 10.0 min residence time. The reaction stream was then collected in a 500 mL flat-bottomed conical flask and concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 3:1) to afford **6b** (2.01 g, 82%, two steps) as a white solid. Mp = 142–143 °C;  $[\alpha]_D^{25} = -22.4$  $(c = 0.28, \text{EtOH}); {}^{1}\text{H NMR (400 MHz, CD}_{3}\text{OD}): \delta 7.93 (d, J = 8.4 \text{ Hz},$ 2H), 7.65 (d, J = 8.4 Hz, 2H), 4.76 (d, J = 5.6 Hz, 1H), 3.53 (dd,  $J_1 =$ 10.8 Hz,  $J_2 = 5.2$  Hz, 1H), 3.39 (dd,  $J_1 = 10.8$  Hz,  $J_2 = 6.0$  Hz, 1H), 3.12 (s, 3H), 2.90 (dd,  $J_1 = 10.8$  Hz,  $J_2 = 5.2$  Hz, 1H) ppm;  $^{13}C\{^1H\}$  NMR  $(100 \text{ MHz}, \text{CD}_3\text{OD})$ :  $\delta$  150.8, 140.7, 128.4, 128.1, 74.0, 63.8, 59.5, 44.2 ppm. ESI HRMS: calcd. for C<sub>10</sub>H<sub>15</sub>NO<sub>4</sub>S + Na 268.0619, found 268.0614.

Continuous Flow Synthesis of (+)-Thiamphenicol (3). For the flow amidation reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume) and a T-mixer. syn-2-Amino-1,3-diol 6b (1.23 g, 5.0 mmol, 1.0 equiv) was dissolved in degassed MeOH (14 mL) and placed in one syringe. The mixture containing methyl dichloroacetate 10a (2.15 g, 15.0 mmol, 3.0 equiv), Et<sub>3</sub>N (1.0 g, 10.0 mmol, 2.0 equiv), and degassed MeOH (3 mL) was placed in another syringe. The syn-2-amino-1,3-diol 6b syringe was set to 77.6  $\mu$ L/min and the methyl dichloroacetate 10a syringe was set to 22.4  $\mu$ L/min. The flow was initiated, and the reaction mixture was heated at 30 °C with a 10.0 min residence time. The reaction stream was collected in a 100 mL flat-bottomed conical flask. The reaction mixture was concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 100:10) to afford (+)-thiamphenicol (3) (1.71 g, 96%) as a white solid. The product was subsequently recrystallized from MeOH/PE, affording (+)-thiamphenicol (3) (1.50 g, 84%). Mp 164–165 °C;  $[\alpha]_D^{20} = +12.6$  (c = 1.0, EtOH); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.90 (d, J = 8.4 Hz, 2H), 7.68  $(d, J = 8.4 \text{ Hz}, 2H), 6.25 \text{ (s, 1H)}, 5.16 \text{ (d, } J = 2.8 \text{ Hz}, 1H), 4.15 \text{ (td, } J_1 = 2.8 \text{ Hz}, 1H), 4.15 \text{ (td, } J_2 = 2.8 \text{ Hz}, 1H), 4.15 \text{ (td, } J_3 = 2.8 \text{$ 6.4 Hz,  $J_2 = 2.8$  Hz, 1H), 3.83 (dd,  $J_1 = 10.8$  Hz,  $J_2 = 6.8$  Hz, 1H), 3.62 (dd,  $J_1 = 10.8 \text{ Hz}$ ,  $J_2 = 6.0 \text{ Hz}$ , 1H), 3.10 (s, 3H) ppm;  $^{13}\text{C}\{^1\text{H}\}$  NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  165.1, 149.0, 139.4, 126.9, 126.8, 70.0, 66.0, 60.8, 57.1, 43.0 ppm. ESI HRMS: calcd. for C<sub>12</sub>H<sub>15</sub>Cl<sub>2</sub>NO<sub>5</sub>S + Na 377.9946, found 377.9941.

((4R,5R)-2-(Dichloromethyl)-5-(4-(methylsulfonyl)phenyl)-4,5-dihydrooxazol-4-yl)methanol (11). For the flow cyclization reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume), a T-mixer, and a 10 bar back-pressure valve. syn-2-Amino-1,3-diol 6b (1.23 g, 5.0 mmol, 1.0 equiv) was dissolved in degassed MeOH (6 mL) and placed in one syringe. The mixture containing dichloroacetonitrile (616 mg, 5.6 mmol, 1.12 equiv), conc. HCl (126 µL), and degassed MeOH (1 mL) was placed in another syringe. The syn-2-amino-1,3-diol 6b syringe was set to 80.6  $\mu$ L/min and the dichloroacetonitrile syringe was set to 19.4  $\mu$ L/min. The flow was initiated, and the reaction mixture was heated at 70 °C with a 10.0 min residence time. The reaction stream was collected in a 100 mL flatbottomed conical flask. The reaction mixture was concentrated in vacuo. The residue was purified by silica gel flash column chromatography (DCM/MeOH = 100:10) to afford 11 (1.49 g, 88%) as a white solid. Mp 144–145 °C;  $[\alpha]_D^{20}$  = +11.2 (c = 1.0, EtOH); <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  7.99 (d, J = 8.0 Hz, 2H), 7.60

(d, J = 8.0 Hz, 2H), 7.26 (s, 1H), 5.75 (d, J = 6.4 Hz, 1H), 5.17 (t, J = 5.6 Hz, 1H), 4.08 (dd, J<sub>1</sub> = 10.8 Hz, J<sub>2</sub> = 5.6 Hz, 1H), 3.75–3.70 (m, 1H), 3.60–3.55 (m, 1H), 3.23 (s, 3H) ppm;  $^{13}$ C{ $^{1}$ H} NMR (100 MHz, DMSO):  $\delta$  161.3, 146.4, 141.0, 128.1, 126.5, 83.3, 76.9, 62.4, 43.9 ppm. ESI HRMS: calcd. for C<sub>12</sub>H<sub>13</sub>Cl<sub>2</sub>NO<sub>4</sub>S + Na 359.9840, found 359.9843.

Continuous Flow Synthesis of (+)-Florfenicol (4), (45,5R)-2-(Dichloromethyl)-4-(fluoromethyl)-5-(4-(methylsulfonyl)phenyl)-4,5-dihydrooxazole (12). For the flow fluorination reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume) and a 10 bar back-pressure valve. Oxazoline 11 (3.38 g, 10.0 mmol, 1.0 equiv) was dissolved in dry DCM (50 mL) and placed in one syringe. The oxazoline 11 syringe was set to 400  $\mu$ L/min. The flow was initiated, and the reaction mixture was heated at 100 °C with a 2.5 min residence time. The reaction stream was collected in a 100 mL flatbottomed conical flask. After collection of the organic reaction solution, the solvent was removed in vacuo to afford product 12, which was used in the next stage without the need for further purification. White solid. Mp 123–126 °C.  $[\alpha]_D^{25} = -12.2$  (c = 0.57, CHCl<sub>3</sub>). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.0 (d, J = 8.0 Hz, 2H), 7.54 (d, J = 8.0 Hz, 2H), 6.35 (s, 1H), 5.72 (d, J = 6.8 Hz, 1H), 4.78-4.53 (m, 2H), 4.35-4.26 (m, 1H), 3.05 (s, 3H) ppm;  ${}^{13}C\{{}^{1}H\}$  NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  163.1, 145.0, 141.0, 128.3, 126.1, 83.2 (d, J = 4.2 Hz), 82.6 (d, J = 173.1 Hz), 74.6 (d, J = 31.9 Hz), 60.9, 44.4 ppm. ESI HRMS: calcd. for  $C_{12}H_{12}Cl_2FNO_3S + Na 361.9797$ , found 361.9795.

(+)-Florfenicol (4). For the flow hydrolysis reaction, the reactor coil consists of a 200 cm segment of a PTFE tube (i.d. = 0.8 mm, 1.0 mL total volume) and a 10 bar back-pressure valve. The above crude 12 was dissolved in  ${}^{i}PrOH/H_{2}O$  (40 mL, v/v = 1.86:1) and placed in one syringe. The syringe was set to 400  $\mu$ L/min. The flow was initiated, and the reaction mixture was heated at 100 °C with a 2.5 min residence time. The reaction stream was collected in a 100 mL flat-bottomed conical flask. The reaction mixture was concentrated in vacuo. The residue was purified by silica gel flash column chromatography (AcOEt/PE = 1:1) to afford (+)-florfenicol (4) (3.26 g, 91%) as a white solid. The product was subsequently recrystallized from AcOEt/ PE, affording (+)-florfenicol (4) (2.90 g, 81%). Mp 151-152 °C;  $[\alpha]_D^{25} = -18.8$  (c = 1.0, DMF); <sup>1</sup>H NMR (400 MHz, DMSO):  $\delta$  8.61 (d, J = 8.8 Hz, 1H), 7.85 (d, J = 8.0 Hz, 2H), 7.61 (d, J = 8.0 Hz, 2H),6.45 (s, 1H), 6.14 (d, J = 4.4 Hz, 1H), 4.98 (t, J = 3.6 Hz, 1H), 4.72-4.57 (m, 1H),  $4.58{-}4.24$  (m, 2H), 3.15 (s, 3H);  $^{13}C\{^{1}H\}$  NMR (100 MHz, DMSO):  $\delta$  164.2, 148.3, 140.0, 127.6, 126.9, 82.8 (d, J = 169.0Hz), 69.8 (d, J = 6.0 Hz), 66.7, 55.0 (d, J = 19.6 Hz), 44.0 ppm. ESI HRMS: calcd. for C<sub>12</sub>H<sub>14</sub>Cl<sub>2</sub>FNO<sub>4</sub>S + Na 379.9902, found 379.9908.

## ASSOCIATED CONTENT

# **Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.1c01124.

Experimental procedures, HPLC data, compound characterization, NMR spectra, and HRMS spectra (PDF)

# AUTHOR INFORMATION

#### **Corresponding Author**

Fener Chen — Sichuan Research Center for Drug Precision Industrial Technology, West China School of Pharmacy, Sichuan University, Chengdu 610041, China; Engineering Center of Catalysis and Synthesis for Chiral Molecules, Department of Chemistry, Fudan University, Shanghai 200433, China; Shanghai Engineering Center of Industrial Asymmetric Catalysis for Chiral Drugs, Shanghai 200433, China; orcid.org/0000-0002-6734-3388; Email: rfchen@fudan.edu.cn

## Authors

Yingqi Xia — Sichuan Research Center for Drug Precision Industrial Technology, West China School of Pharmacy, Sichuan University, Chengdu 610041, China

- Meifen Jiang Engineering Center of Catalysis and Synthesis for Chiral Molecules, Department of Chemistry, Fudan University, Shanghai 200433, China
- Minjie Liu Shanghai Engineering Center of Industrial Asymmetric Catalysis for Chiral Drugs, Shanghai 200433, China
- Yan Zhang Sichuan Research Center for Drug Precision Industrial Technology, West China School of Pharmacy, Sichuan University, Chengdu 610041, China
- Hongmin Qu Sichuan Research Center for Drug Precision Industrial Technology, West China School of Pharmacy, Sichuan University, Chengdu 610041, China
- Tong Xiong Sichuan Research Center for Drug Precision Industrial Technology, West China School of Pharmacy, Sichuan University, Chengdu 610041, China
- Huashan Huang Shanghai Engineering Center of Industrial Asymmetric Catalysis for Chiral Drugs, Shanghai 200433, China
- Dang Cheng Engineering Center of Catalysis and Synthesis for Chiral Molecules, Department of Chemistry, Fudan University, Shanghai 200433, China; orcid.org/0000-0001-5899-5141

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.joc.1c01124

#### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

This work was supported by the Fundamental Research Funds for the Central Universities (YJ201805 and YJ201864).

## REFERENCES

- (1) (a) Cutler, R. A.; Stenger, R. J.; Suter, C. M. New Antibacterial Agents. 2-Acylamino-1-(4-hydrocarbonylsulfonylphenyl)-1,3-propanediols and Related Compounds. J. Am. Chem. Soc. 1952, 74, 5475-5478. (b) Suter, C. M.; Schalit, S.; Cutler, R. A. New Antibacterial Agents. II. An Alternate Synthesis of DL-threo-2-Dichloro-acetamido-1-(4-methylsulfonylphenyl)-1, 3-propanediol. J. Am. Chem. Soc. 1953, 75, 4330-4333. (c) Jommi, G.; Pagliarin, R.; Chiarino, D.; Fantucci, M. 2-Oxazolidinones as regioselective protection of  $\beta$ -amino alcohols in the synthesis of 2-amino-1-aryl-3-fluoro-1-propanols. Gazz. Chim. Ital. 1985, 115, 653-658. (d) Schumacher, D. P.; Clark, J. E.; Murphy, B. L.; Fischer, P. A. An efficient synthesis of florfenicol. J. Org. Chem. 1990, 55, 5291-5294. (e) Giordano, C.; Cavicchioli, S.; Levi, S.; Villa, M. Direct conversion of (1S,2S)-2-amino-1-[(4-methylthio)phenyl]-1,3propanediol into its enantiomer for efficient synthesis of thiamphenicol and florfenicol. J. Org. Chem. 1991, 56, 6114-6118. (f) Clark, J. E.; Fischer, P. A.; Schumacher, D. P. An Enzymatic Route to Florfenicol. Synthesis 1991, 1991, 891-894.
- (2) (a) Horsberg, T. E.; Hoff, K. A.; Nordmo, R. Pharmacokinetics of Florfenicol and Its Metabolite Florfenicol Amine in Atlantic Salmon. *J. Aquat. Anim. Health* **1996**, *8*, 292–301. (b) Shen, J.; Hu, F.; Wu, X.; Coats, J. R. Bioavailability and pharmacokinetics of florfenicol in broiler chickens. *J. Vet. Pharmacol. Ther.* **2003**, *26*, 337–341. (c) Ueda, Y.; Ohtsuki, S.; Narukawa, N. Efficacy of Florfenicol on Experimental Actinobacillus Pleuropneumonia in Pigs. *J. Vet. Med. Sci.* **1995**, *57*, 261–265. (d) Ehrlich, J.; Bartz, Q. R.; Smith, R. M.; Joslyn, D. A.; Burkholder, P. R. Chloromycetin, a New Antibiotic From a Soil Actinomycete. *Science* **1947**, *106*, 417. (e) Rebstock, M. C.; Crooks, H. M.; Controulis, J.; Bartz, Q. R. Chloramphenicol (Chloromycetin).1 IV.1a Chemical Studies. *J. Am. Chem. Soc.* **1949**, *71*, 2458–2462.
- (3) (a) Madhavan, H. N.; Bagyalakshmi, R. Farewell, Chloramphenicol? Is this True?: A Review. *Res. Rev. J. Microbiol. Biotechnol.* **2014**, *3*, 13–26. (b) Aspa, J.; Rajas, O.; Castro, F. R.; Blanquer, J.; Zalacain, R.;

- Fenoll, A.; Celis, R.; Vargas, A.; Salvanes, F. R.; Espana, P. P.; Rello, J.; Torres, A. Drug-Resistant Pneumococcal Pneumonia: Clinical Relevance and Related Factors. Clin. Infect. Dis. 2004, 38, 787-798. (c) Campos, J.; Román, F.; Pérez-Vázquez, M.; Aracil, B.; Oteo, J.; Cercenado, E. Antibiotic resistance and clinical significance of Haemophilus influenzae type f. J. Antimicrob. Chemother. 2003, 52, 961–966. (d) Duke, T. Neonatal pneumonia in developing countries. Arch. Dis. Child. Fetal Neonatal Ed. 2005, 90, F211-F219. (e) Hartmann, C.; Peter, C.; Hermann, E.; Ure, B.; Sedlacek, L.; Hansen, G.; Bohnhorst, B. Successful treatment of vancomycinresistant Enterococcus faecium ventriculitis with combined intravenous and intraventricular chloramphenicol in a newborn. J. Med. Microbiol. 2010, 59, 1371–1374. (f) Sood, S.; Kapil, A.; Das, B.; Jain, Y.; Kabra, S. K. Re-emergence of chloramphenicolsensitive Salmonella typhi. Lancet 1999, 353, 1241-1242. (g) Sowmiya, M.; Malathi, J.; Madhavan, H. N. Screening of Ocular Enterobacteriaceae Isolates for Presence of Chromosomal blaNDM-1 and ESBL Genes: A 2-Year Study at a Tertiary Eye Care Center. Invest. Ophthalmol. Vis. Sci. 2012, 53, 5251-5257. (h) Du, Z.; Wang, M.-Y.; Cui, G.-Y.; Zu, X.-Y.; Zhao, Z.-Q.; Xue, Y. The prevalence of amphenicol resistance in Escherichia coli isolated from pigs in mainland China from 2000 to 2018: A systematic review and meta-analysis. PLoS One 2020, 15, No. e0228388.
- (4) Nitzan, O.; Supnitzky, U.; Kennes, Y.; Chazan, B.; Raul, R.; Colodner, R. Is Chloramphenicol Making a Comeback? *Isr. Med. Assoc. J.* **2010**, *12*, 371–374.
- (5) (a) Corey, E. J.; Choi, S. Efficient Enantioselective Syntheses of Chloramphenicol and (D)-threo- and (D)-erythro-Sphingosine. Tetrahedron Lett. 2000, 41, 2765-2768. (b) Hajra, S.; Karmakar, A.; Maji, T.; Medda, A. K. Stereoselective syntheses of (-)-chloramphenicol and (+)-thiamphenicol. Tetrahedron 2006, 62, 8959-8965. (c) Loncaric, C.; Wulff, W. D. An Efficient Synthesis of (-)-Chloramphenicol via Asymmetric Catalytic Aziridination: A Comparison. Org. Lett. 2001, 3, 3675-3678. (d) Bhaskar, G.; Kumar, V. S.; Rao, B. V. A Short Stereoselective Synthesis of (-)-Chloramphenicol and (+)-Thiamphenicol. Tetrahedron: Asymmetry 2004, 15, 1279-1283. (e) Wu, G.; Schumacher, D. P.; Tormos, W.; Clark, J. E.; Murphy, B. L. An Improved Industrial Synthesis of Florfenicol plus an Enantioselective Total Synthesis of Thiamphenicol. J. Org. Chem. 1997, 62, 2996–2998. (f) Franchino, A.; Jakubec, P.; Dixon, D. J. Enantioselective Synthesis of (-)-Chloramphenicol via Silver-Catalysed Asymmetric Isocyanoacetate Aldol Reaction. Org. Biomol. Chem. 2016, 14, 93-96. (g) Perez, M.; Echeverria, P.-G.; Martine-Arripe, E.; Zoubir, M. E.; Touati, R.; Zhang, Z.; Genet, J.-P.; Phansavath, P.; Ayad, T.; Ratovelomanana-Vidal, V. An Efficient Stereoselective Total Synthesis of All Stereoisomers of the Antibiotic Thiamphenicol through Ruthenium-Catalyzed Asymmetric Reduction by Dynamic Kinetic Resolution. Eur. J. Org. Chem. 2015, 2015, 5949-5958. (h) Zou, J.; Ni, G.; Tang, J.; Yu, J.; Jiang, L.; Ju, D.; Zhang, F.; Chen, S. Asymmetric Synthesis of Florfenicol by Dynamic Reductive Kinetic Resolution with Ketoreductases. Eur. J. Org. Chem. 2018, 2018, 5044-5053. (i) Lu, W.; Chen, P.; Lin, G. New Stereoselective Synthesis of Thiamphenicol and Florfenicol from Enantiomerically Pure Cyanohydrin: a Chemo-Enzymatic Approach. Tetrahedron 2008, 64, 7822-7827.
- (6) Li, F.; Wang, Z.-H.; Zhao, L.; Chen, F.-E. A Facile and Efficient Asymmetric Synthesis of Florfenicol. *Synlett* **2011**, *19*, 2883–2885.
- (7) Li, F.; Wang, Z.-H.; Zhao, L.; Xiong, F.-J.; He, Q.-Q.; Chen, F.-E. An efficient enantioselective synthesis of florfenicol via a vanadium-catalyzed asymmetric epoxidation. *Tetrahedron: Asymmetry* **2011**, 22, 1337–1341.
- (8) Wang, Z.-H.; Zheng, C.; Li, F.; Zhao, L.; Chen, F.-E.; He, Q.-Q. An Efficient Enantioselective Synthesis of Florfenicol Based on Sharpless Asymmetric Dihydroxylation. *Synthesis* **2012**, *44*, 699–704.
- (9) Wang, X.-L.; Xu, L.-J.; Yan, L.-J.; Wang, H.-F.; Han, S.; Wu, Y.; Chen, F.-E. Catalytic asymmetric transfer hydrogenation/dynamic kinetic resolution: an efficient synthesis of florfenicol. *Tetrahedron* **2016**, 72, 1787–1793.
- (10) Liu, J.-X.; Li, Y.-L.; Ke, M.-L.; Liu, M.-J.; Zhan, P.-P.; Xiao, Y.-C.; Chen, F.-E. Unified Strategy to Amphenical Antibiotics: Asymmetric Synthesis of (-)-Chloramphenical, (-)-Azidamphenical, and

- (+)-Thiamphenicol and Its (+)-3-Floride. J. Org. Chem. 2020, 85, 15360–15367.
- (11) Wang, Z.-H.; Li, F.; Zhao, L.; He, Q.-Q.; Chen, F.-E.; Zheng, C. An efficient enantioselective synthesis of florfenicol via asymmetric aziridination. *Tetrahedron* **2011**, *67*, 9199–9203.
- (12) (a) Kim, Y.; Li, C.-J. Perspectives on green synthesis and catalysis. *Green Synth. Catal.* **2020**, *1*, 1–11. (b) Trost, B. M. On Inventing Reactions for Atom Economy. *Acc. Chem. Res.* **2002**, *35*, 695–705. (c) Sheldon, R. A. Metrics of Green Chemistry and Sustainability: Past, Present, and Future. *ACS Sustainable Chem. Eng.* **2018**, *6*, 32–48. (d) Li, L.; Chen, Z.; Zhang, X.-W.; Jia, Y.-X. Divergent Strategy in Natural Product Total Synthesis. *Chem. Rev.* **2018**, *118*, 3752–3832.
- (13) (a) Qin, D.-D.; Lai, W.-H.; Hu, D.; Chen, Z.; Wu, A.-A.; Ruan, Y.-P.; Zhou, Z.-H.; Chen, H.-B. Highly Enantioselective Henry Reactions of Aromatic Aldehydes Catalyzed by an Amino Alcohol—Copper(II) Complex. *Chem. Eur. J.* **2012**, *18*, 10515—10518. (b) Qin, D.-D.; Yu, W.; Zhou, J.-D.; Zhang, Y.-C.; Ruan, Y.-P.; Zhou, Z.-H.; Chen, H.-B. syn- and Enantioselective Henry Reactions of Aliphatic Aldehydes and Application to the Synthesis of Safingol. *Chem. Eur. J.* **2013**, *19*, 16541—16544.
- (14) For reviews on the asymmetric Henry reactions: (a) Luzzio, F. A. The Henry reaction: recent examples. *Tetrahedron* **2001**, *57*, 915–945. (b) Palomo, C.; Oiarbide, M.; Mielgo, A. Unveiling Reliable Catalysts for the Asymmetric Nitroaldol (Henry) Reaction. Angew. Chem., Int. Ed. 2004, 43, 5442-5444. (c) Palomo, C.; Oiarbide, M.; Laso, A. Recent Advances in the Catalytic Asymmetric Nitroaldol (Henry) Reaction. Eur. J. Org. Chem. 2007, 2007, 2561-2574. (d) Murugavel, G.; Sadhu, P.; Punniyamurthy, T. Copper(II)-Catalyzed Nitroaldol (Henry) Reactions: Recent Developments. Chem. Rec. 2016, 16, 1906-1917. (e) Saranya, S.; Harry, N. A. S.; Ujwaldev, M.; Anilkumar, G. Recent Advances and Perspectives on the Zinc-Catalyzed Nitroaldol (Henry) Reaction. Asian J. Org. Chem. 2017, 6, 1349-1360. (f) Zhang, S.; Li, Y.-N.; Xu, Y.-G.; Wang, Z.-Y. Recent progress in copper catalyzed asymmetric Henry reaction. Chin. Chem. Lett. 2018, 29, 873-883. (g) Dong, L.; Chen, F.-E. Asymmetric catalysis in direct nitromethanefree Henry reactions. RSC Adv. 2020, 10, 2313-2326. (h) Boruwa, J.; Gogoi, N.; Saikia, P. P.; Barua, N. C. Catalytic asymmetric Henry reaction. Tetrahedron: Asymmetry 2006, 17, 3315-3326. (i) Marcelli, T.; Haas, R. N. S.; Maarseveen, J. H.; Hiemstra, H. Asymmetric Organocatalytic Henry Reaction. Angew. Chem., Int. Ed. 2006, 45, 929-931.
- (15) (a) Tseng, S.-L.; Yang, T.-K. New β-amino thiols as efficient catalysts for highly enantioselective alkenylzinc addition to aldehydes. *Tetrahedron: Asymmetry* **2005**, *16*, 773–782. (b) Guo, Z.-L.; Zhong, S.; Li, Y.-B.; Lu, G. Chiral 1,1′-binaphthylazepine derived amino alcohol catalyzed asymmetric Henry reaction. *Tetrahedron: Asymmetry* **2011**, 22, 238–245.
- (16) For reviews on flow chemistry: (a) Pastre, J. C.; Browne, D. L.; Ley, S. V. Flow chemistry syntheses of natural products. Chem. Soc. Rev. 2013, 42, 8849-8869. (b) Plutschack, M. B.; Pieber, B.; Gilmore, K.; Seeberger, P. H. The Hitchhikers Guide to Flow Chemistry. Chem. Rev. 2017, 117, 11796-11893. (c) Wiles, C.; Watts, P. Recent advances in micro reaction technology. Chem. Commun. 2011, 47, 6512-6535. (d) Baumann, M.; Baxendale, I. R. The synthesis of active pharmaceutical ingredients (APIs) using continuous flow chemistry. Beilstein J. Org. Chem. 2015, 11, 1194-1219. (e) Porta, R.; Benaglia, M.; Puglisi, A. Flow Chemistry: Recent Developments in the Synthesis of Pharmaceutical Products. Org. Process: Res. Dev. 2016, 20, 2-25. (f) Rogers, L.; Jensen, K. F. Continuous manufacturing - the Green Chemistry promise? Green Chem. 2019, 21, 3481-3498. (g) Gutmann, B.; Cantillo, D.; Kappe, C. O. Continuous-Flow Technology-A Tool for the Safe Manufacturing of Active Pharmaceutical Ingredients. Angew. Chem., Int. Ed. 2015, 54, 6688-6728. (h) Jas, G.; Kirschning, A. Continuous Flow Techniques in Organic Synthesis. Chem. - Eur. J. 2003, 9, 5708-5723. (i) Liao, J.-Y.; Zhang, S.-L.; Wang, Z.-S.; Song, X.; Zhang, D.-L.; Kumar, R.; Jin, J.; Ren, P.; You, H.-Z.; Chen, F.-E. Transition-metal catalyzed asymmetric reactions under continuous flow from 2015 to early 2020. Green Synth. Catal. 2020, 1, 121-133.

(17) For selected review and example on hydrogenation in continuous-flow: (a) Irfan, M.; Glasnov, T. N.; Kappe, C. O. Heterogeneous Catalytic Hydrogenation Reactions in Continuous-Flow Reactors. ChemSusChem 2011, 4, 300-316. (b) Newton, S.; Ley, S. V.; Arce, E. C.; Grainger, D. M. Asymmetric Homogeneous Hydrogenation in Flow using a Tube-in-Tube Reactor. Adv. Synth. Catal. 2012, 354, 1805-1812. (c) Ishitani, H.; Kanai, K.; Yoo, W.-J.; Yoshida, T.; Kobayashi, S. Nickel-Diamine/Mesoporous Silica Composite as a Heterogeneous Chiral Catalyst for Asymmetric 1,4-Addition Reactions. Angew. Chem., Int. Ed. 2019, 58, 13313-13317. (d) Yoswathananont, N.; Nitta, K.; Nishiuchi, Y.; Sato, M. Continuous hydrogenation reactions in a tube reactor packed with Pd/C. Chem. Commun. 2005, 40-42. (e) Saaby, S.; Knudsen, K. R.; Ladlowb, M.; Ley, S. V. The use of a continuous flow-reactor employing a mixed hydrogen-liquid flow stream for the efficient reduction of imines to amines. Chem. Commun. 2005, 2909-2911. (f) Yang, C.-X.; Teixeira, A. R.; Shi, Y.-X.; Born, S. C.; Lin, H.-K.; Song, Y. F. L.; Martin, B.; Schenkel, B.; Lachegurabia, M. P.; Jensen, K. F. Catalytic hydrogenation of N-4-nitrophenyl nicotinamide in a micro-packed bed reactor. Green Chem. 2018, 20, 886-893. (g) Tu, J.-C.; Sang, L.; Cheng, H.; Ai, N.; Zhang, J.-S. Continuous Hydrogenolysis of N-Diphenylmethyl Groups in a Micropacked-Bed Reactor. Org. Process: Res. Dev. 2020, 24, 59-66.